

# Your thoughts/comments/hopes/wishes

Three physicists held a beach party and had so much fun they made it annual. Its now known as a popular wave function among the department.

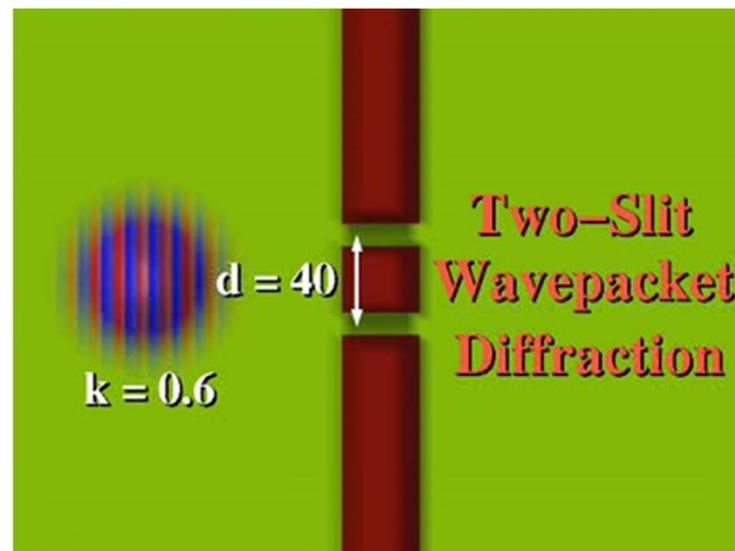
Can we go over some worked examples of applying wave equations like in the checkpoint?

I feel like im getting my math wrong

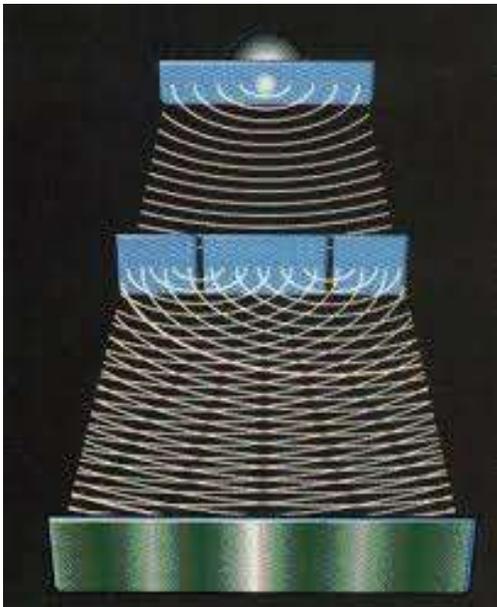
Yep, we'll go over it. It's not too bad once you get used to it. Like weather in Illinois.

I tried to think of a quantum mechanics joke, but the thing about them is that they can be incredibly funny and incredibly unfunny at the same time. 😊

# Lecture 6: The Wave Function



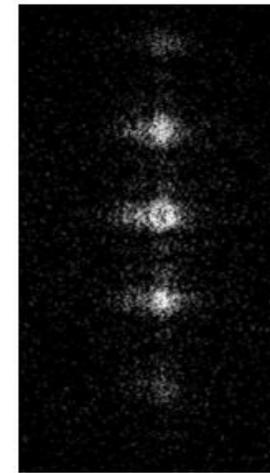
# Last time: Interference of individual photons



5 photons



150 photons



15,000 photons

*Low intensity interference experiment using a single photon counting camera. The photons first appear to arrive at random positions, but after many photons have arrived an interference pattern emerges.*

# Double slit experiment for electrons

Suppose that, instead of photons, we electrons one at a time through a pair of slits. We place an electron detector on a distant screen. Do we expect to see interference?

- a) No, interference is due to electrons interfering with one another.
- b) No, interference is a wave property while electrons are particles.
- c) Yes, the electron's wave function will have interference.

# Wave function summary

$\psi(x, t)$  is a complex number.

The probability density is given by

$$\rho(x, t) = |\psi(x, t)|^2 = \psi(x, t)\psi^*(x, t)$$

The probability to find the particle between  $a$  and  $b$  at time  $t$  is

$$P(a < x < b, t) = \int_a^b \rho(x, t) dx$$

# Checkpoint

An electron has the wave function  $\frac{1}{\sqrt{2}}(\psi_1 + \psi_2)$ , where

$$\psi_1 = \frac{1}{\sqrt{2}} + \frac{i}{\sqrt{2}} \text{ for } -2 < x \leq -1,$$

$$\psi_2 = 1 \text{ for } 1 < x \leq 2,$$

And zero elsewhere. If we measure the position of the electron, what is the probability we find it with  $x \leq 0$ ?

- a) 0
- b)  $\frac{1}{4}$
- c)  $\frac{1}{2}$
- d)  $\frac{3}{4}$
- e) 1

# Normalization

Suppose that  $\psi(x) = A \sin \frac{2\pi x}{L}$  for  $0 < x < L$ , and 0 otherwise. What equation must A satisfy?

a)  $A \int_{-\infty}^{\infty} \sin \frac{2\pi x}{L} dx = 1$

b)  $A \int_0^L \sin \frac{2\pi x}{L} dx = 1$

c)  $|A|^2 \int_{-\infty}^{\infty} \left| \sin \frac{2\pi x}{L} \right|^2 dx = 1$

d)  $|A|^2 \int_0^L \left| \sin \frac{2\pi x}{L} \right|^2 dx = 1$

e)  $|A|^2 = 1$

# Wave functions and probabilities

Suppose that we are told that the wave function for an electron is given by

$$\psi(x) = \frac{1}{\sqrt{2}}(1 + i)$$

$\text{nm}^{-1/2}$  between  $x=2 \text{ nm}$  and  $x=2.5 \text{ nm}$ , where there is a sensor that will register a 'click' if an electron is detected.



What is the probability that the sensor will click?

- a) 0
- b)  $\frac{1}{4}$
- c)  $\frac{1}{2}$
- d)  $\frac{3}{4}$
- e) 1

# What is the wave function of a particle with wavelength $\lambda$ ?

What we know:

Must interfere like classical light with wavelength  $\lambda$ .

Probability must be proportional to the intensity computed classically.

A guess:

$$\Psi(x, t) \propto e^{i(kx - \omega t)}$$

# What if particles were like photons

Particle with momentum  $p = \frac{h}{\lambda} = \hbar k$ :

$$\psi(x, t) = A e^{i(kx - \omega t)}.$$

Superposition of paths:

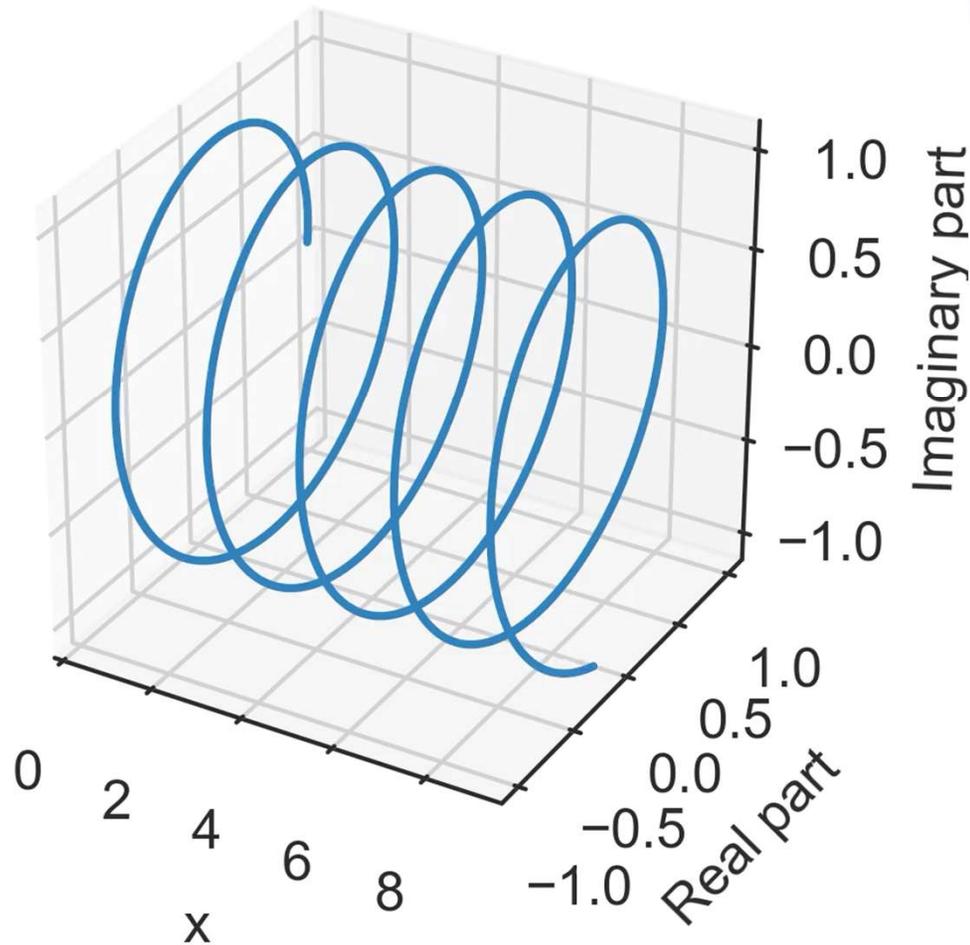
$$\psi(y, t) = A(e^{ikr_1} + e^{ikr_2})e^{i\omega t}$$

$$\rho(y, t) = |A|^2 |e^{ikr_1} + e^{ikr_2}|^2$$

deBroglie hypothesis!

If this is true, then passing particles through slits would result in interference fringes in the probability that they arrive at a screen.

# Wave function of a particle with momentum $p$



Empirical fact:

$$E = hf = \hbar\omega$$

$$p = \frac{h}{\lambda} = \hbar k$$

$$\psi(x, t) = Ae^{i(kx - \omega t)}$$

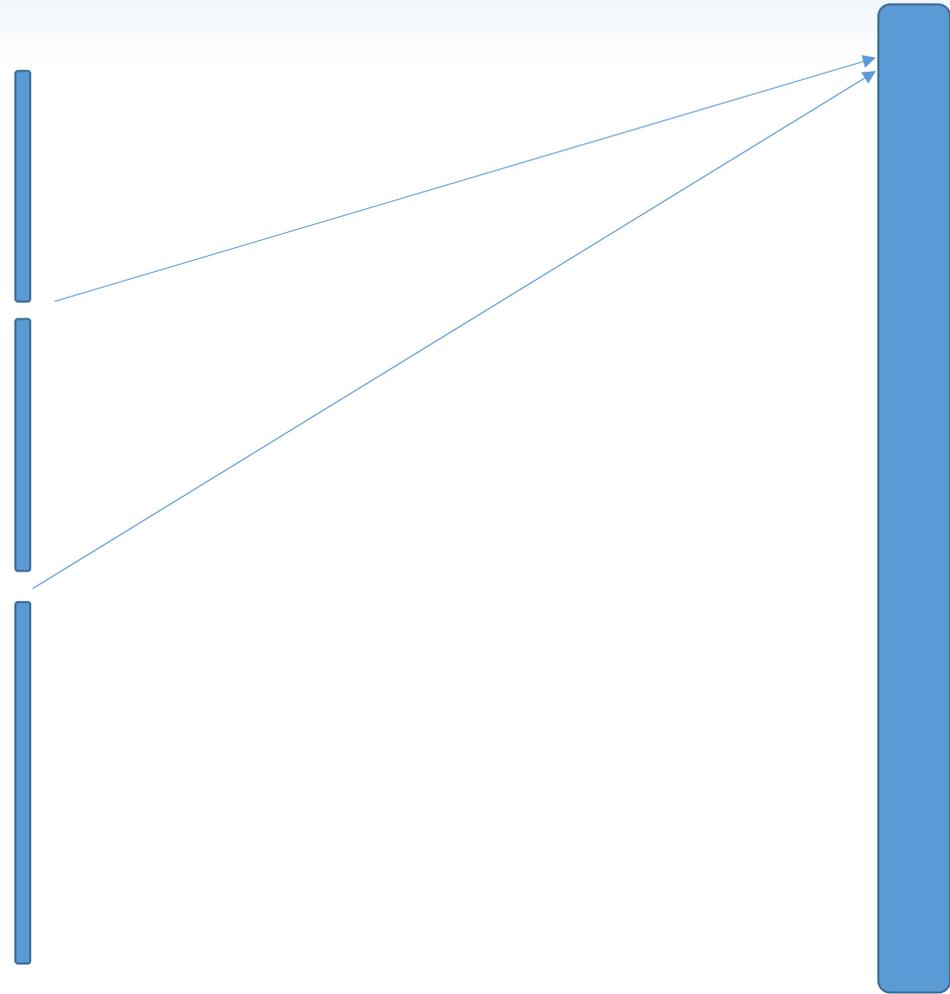
# The quantum description of the two-slit experiment

Particle with wavelength  $\lambda = \frac{2\pi}{k}$

$$\psi(x, t) = Ae^{i(kx - \omega t)}$$

Superposition of paths:

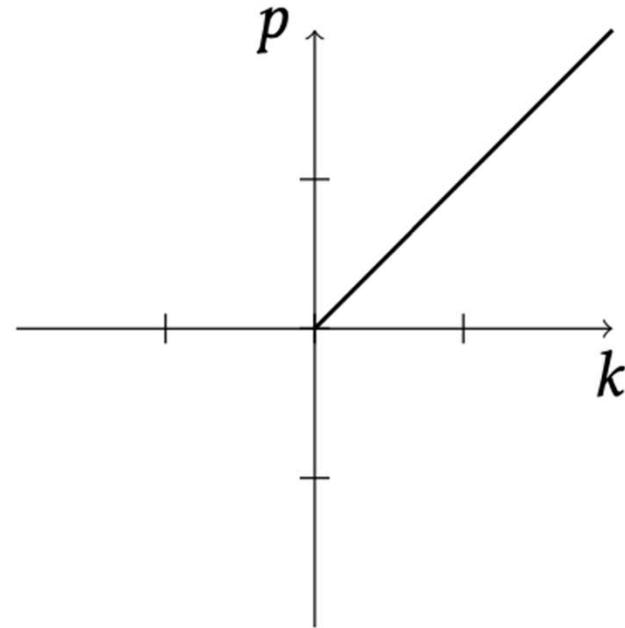
$$\psi(y, t) = A(e^{ikr_1} + e^{ikr_2})e^{i\omega t}$$



$$p = \hbar k$$

Relationship between momentum  
and wave number is empirical!  
(based on experiment)

But it is the same relationship for  
electrons and for photons.

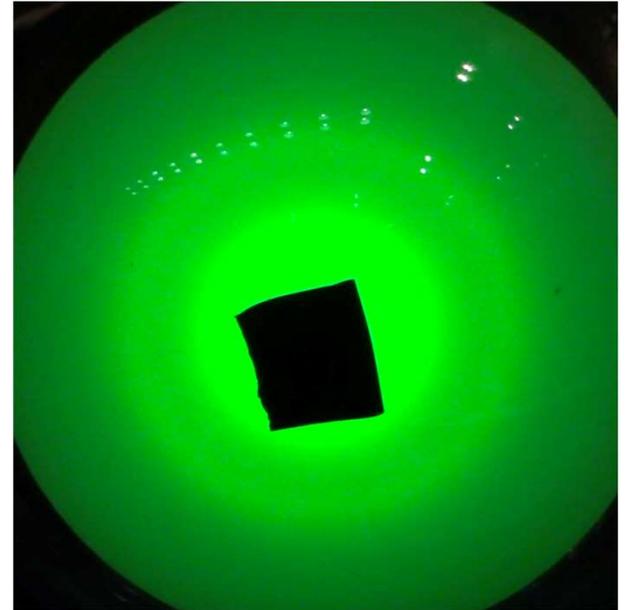


# Measuring momentum

Suppose that electrons with momentum  $p$  are described by a wave function  $e^{ikx}$ , with  $p = \hbar k$ .

If we decrease the velocity of the electrons, what will happen to the spacing between fringes?

- a) Decrease.
- b) Stay the same.
- c) Increase.



# Final summary

$\psi(x, t)$  is a complex number.

The probability density is given by

$$\rho(x, t) = |\psi(x, t)|^2$$

The probability to find the particle between  $a$  and  $b$  at time  $t$  is

$$P(a < x < b, t) = \int_a^b \rho(x, t) dx$$

Superposition: Add wave functions, then square to get probabilities (like waves)

# Review

Light comes in packets called photons with energy  $hf$  and momentum  $h/\lambda$ .

We describe quantum objects using a wave function  $\psi(x, t)$ . The probability density that the object is observed at  $x, t$  is  $\psi(x, t)\psi^*(x, t)$ .

Quantum objects like electrons also have a wavelength given by  $\lambda = h/mv$

# Adding waves of unequal amplitude

Suppose that the total wave function of an electron at a given spot on the screen (within a small region)  $x$  is given by  $\psi(x) = 2e^{ikr_1} + 3e^{ikr_2} \text{ m}^{-1/2}$ .

What is the maximum probability density that the electron will be observed at that spot?

- a)  $0.5 \text{ m}^{-1}$
- b)  $1 \text{ m}^{-1}$
- c)  $5 \text{ m}^{-1}$
- d)  $25 \text{ m}^{-1}$

What's the minimum probability density that the electron will be observed at that spot?

- a)  $0.5 \text{ m}^{-1}$
- b)  $1 \text{ m}^{-1}$
- c)  $5 \text{ m}^{-1}$
- d)  $25 \text{ m}^{-1}$