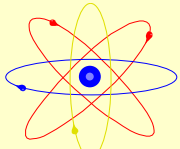


Lect. 19 - Big Picture: Smallest objects to the Universe


The Big Picture

The smallest objects to the Universe

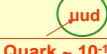
Atom - size $\sim 10^{-10}$ m
Nucleus - size $\sim 10^{-14}$ m



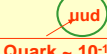
Person - size ~ 1 m



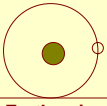

Proton - size $\sim 10^{-15}$ m



Quark $\sim 10^{-18}$ m



Earth - size $\sim 10^7$ m
Solar System - size $\sim 10^{11}$ m

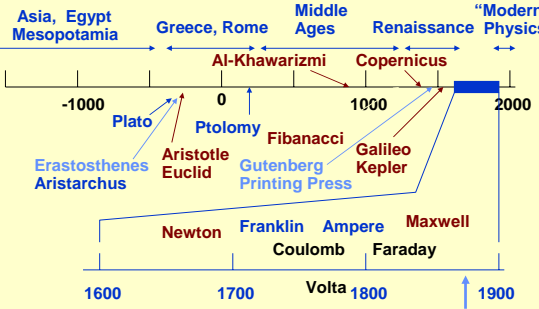



Universe - size $\sim 10^{26}$ m

Announcements

- **Schedule:**
 - **Today:** Start the conclusion of the course -- Examples of modern physics
The Big Picture: Powers of Ten: Enormous ranges of sizes and magnitudes in science
 - **Today:** Introduction to ideas of Quantum mechanics
The second revolution of the 20th century
Essential to understand nature at the small scale
 - **Next Time:** Discovery of the electron, nucleus – origins of quantum mechanics
March (Ch 13-14-15)
- **REPORT**
 - Outline for your report / essay due Mon. November 17
 - See "suggestions of topics" on www pages
Today – outline of last part of course to help you decide on a topic

Timeline

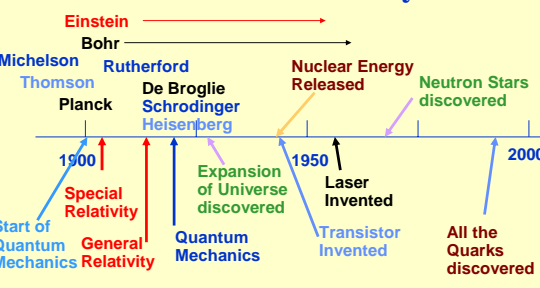


Asia, Egypt Mesopotamia | Greece, Rome | Middle Ages | Renaissance | "Modern" Physics

-1000: Eratosthenes, Aristotle, Euclid, Ptolemy, Al-Khwarizmi, Fibonacci, Copernicus, Galileo, Kepler, Newton, Franklin, Ampere, Maxwell, Coulomb, Faraday, Volta, 1800, 1900

- "Classical Physics" was complete around 1880
- See Timeline description of lives of various scientists on WWW pages.

Timeline - Modern Physics



1900: Start of Quantum Mechanics, Special Relativity, General Relativity, Bohr, Planck, Thomson, Rutherford, De Broglie, Schrodinger, Heisenberg, Einstein

1950: Expansion of Universe discovered, Laser Invented, Transistor Invented, Nuclear Energy Released

2000: Neutron Stars discovered, All the Quarks discovered

- "Modern Physics" was a sudden revolution starting around 1900, and ending ????
- See Timeline description of lives of various scientists on WWW pages.

Objects in our universe

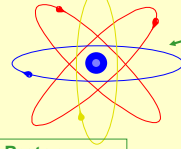
Enormous range of sizes

- **Classical Mechanics** works very well for **ordinary objects** around us:
 - Scale of size ~ 1 m -- From the smallest object you can see $\sim 10^{-4}$ m to Planets of order 10^7 m
 - Typical velocities ~ 100 m/s (fastest rockets $\sim 10^4$ m/s)
- **Relativity** (Special and General) are important for objects moving **very fast** or **very massive**
 - Speeds near $c = 10^8$ m/s
 - Protons, Neutrons inside nuclei have very high energies
 - Important for motions of galaxies at large fractions of c
 - Strong effects of gravity -- black holes
- **Quantum Mechanics** crucial to understand the **very small** units from which matter is made
 - Typical sizes: Atoms, Nuclei, ... ($\sim 10^{-10}$ - 10^{-14} m)

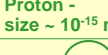
How do we know the sizes of atoms?

What they are made of?

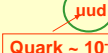
Atom - size $\sim 10^{-10}$ m
Nucleus - size $\sim 10^{-14}$ m



Proton - size $\sim 10^{-15}$ m



Quark $\sim 10^{-18}$ m



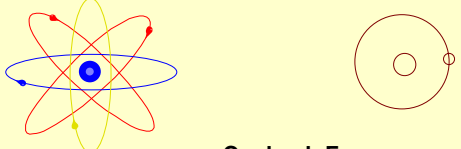
Next lecture
Discovery of electron and nucleus led to basic picture
Led to Quantum Mechanics
Uncertainty Principle
Many important inventions
Lasers
Semiconductors (Your computer!)
....

Later
Understanding of the Universe!

Lect. 19 - Big Picture: Smallest objects to the Universe

Ratios of different forces Enormous range of magnitudes

Atom - Electrical Forces Solar System - Gravitational
Hold electrons to nucleus Forces hold earth to sun

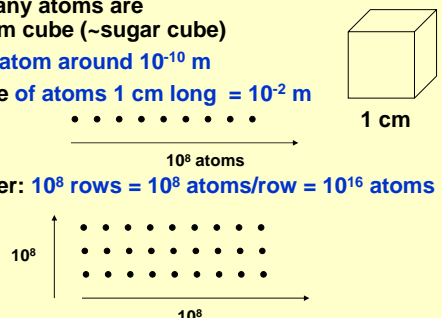


For electrons: $\frac{\text{Coulomb Force}}{\text{Force of Gravity}} \sim 10^{41}$

Forces inside nucleus **MUCH** stronger than Coulomb Forces! (More later)

Example of enormous range of sizes

- How many atoms are in a 1 cm cube (~sugar cube)
- Size of atom around 10^{-10} m
- One line of atoms 1 cm long = 10^8 m



One layer: 10^8 rows = 10^8 atoms/row = 10^{16} atoms

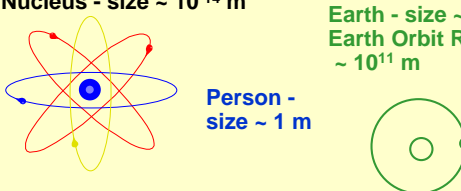
Cube: 10^8 layers x 10^{16} atoms/layer = 10^{24} atoms

Ratios of sizes

Atom - size $\sim 10^{-10}$ m
Nucleus - size $\sim 10^{-14}$ m

Earth - size $\sim 10^7$ m
Earth Orbit Radius $\sim 10^{11}$ m

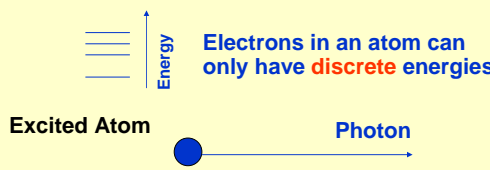
Person - size ~ 1 m



$\frac{\text{Size of atom}}{\text{Size of person}} \sim \frac{\text{Size of person}}{\text{Size of radius of earth orbit}}$

Quantum Effects Crucial for Small Objects

Example - electrons in atoms



Electrons in an atom can only have **discrete** energies


Excited Atom → Photon

Light is emitted only at **discrete** energies, i.e., **discrete** frequencies for each type of atoms

Quantum mechanics explains the stability of the atom -- why the electron does not "fall" into the nucleus

Quantum Mechanics: Particles Act Like Waves!

Prince Louis De Broglie
Matter Waves

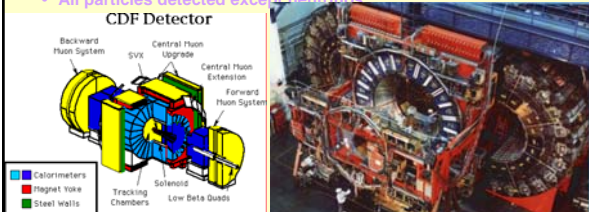


Heisenberg's **Uncertainty Principle**

Schrodinger's **Equation**

Search for the smallest particles

- Experiment at Fermilab (near Chicago) to detect what is produced when high energy (900 GeV) protons and anti-protons collide.
- Energies of particles determined by energy deposition in calorimeter.
- All particles detected except neutrinos



CDF Detector

Backward Photon System, Central Photon Upgrade, Central Photon Extension, Forward Photon System, Calorimeters, Magnet Yoke, Steel walls, Tracking Chambers, Solenoid, Low Beta Guide

Lect. 19 - Big Picture: Smallest objects to the Universe

Important Quantum Effects in Our World
I Lasers

Usually light is emitted by an excited atom in a random direction - light from many atoms goes in all directions

Excited Atoms Photons

What is special about a **Laser**??

Important Quantum Effects in Our World
I Lasers

Lasers work because of the **quantum properties of photons** -- one photon tends to cause another to be emitted

One Photon Excited Atoms Many Photons

If there are many excited atoms, the photons can "cascade" -- very intense, collimated light is emitted forming a beam of precisely the same color light

Important Quantum Effects in Our World
Electrons in crystals

Energy

Electrons in an atom can have only discrete "allowed" energies with "forbidden gaps"

Allowed energies

Forbidden energies

Electrons in a crystal can have bands of "allowed" energies with "forbidden gaps"

Due to Wave character of electrons - Interference!

Important Quantum Effects in Our World
Semiconductors
The basis of all modern electronics
 Transistor invented at Bell Labs, 1947
 (Bardeen, Brattain, Shockley)

In semiconductors the active extra electrons go here

Forbidden energies

Energy

Electrons in a crystal can have bands of "allowed" energies with "forbidden gaps"

Due to Wave character of electrons - Interference!

Important Quantum Effects in Our World
Superconductivity
 Discovered in 1911 by K. Onnes
 Completely baffling in classical physics

Current flowing without loss -- flows forever!

wire

Explained in 1957 by Bardeen, Cooper And Shrieffer at the Univ. of Illinois. (Bardeen is the only person to win two Nobel Prizes in the same field!)
 Due to all the electrons acting together to form a single quantum state -- electrons flow around a wire like the electrons in an atom!

Demonstration
"High - Temperature Superconductors"
 Discovered in 1987 (Nobel Prize)
 (Still not understood!)

Superconductor levitated above magnet - repelled due to currents in superconductor

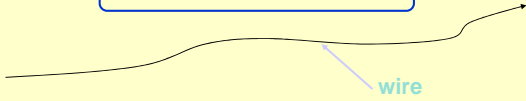
Magnet

Lect. 19 - Big Picture: Smallest objects to the Universe

Important Quantum Effects in Our World
Superconductivity

Completely baffling in classical physics

Current flowing without loss



wire

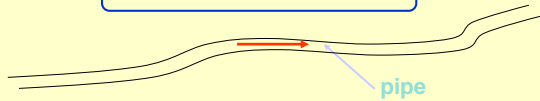
Electric Power lines could carry electricity from California to New York with **no loss of power!**

Possible now, but not economically feasible

Quantum Effects in Our World
Superfluids

Completely baffling in classical physics

Fluids that flow without loss



pipe

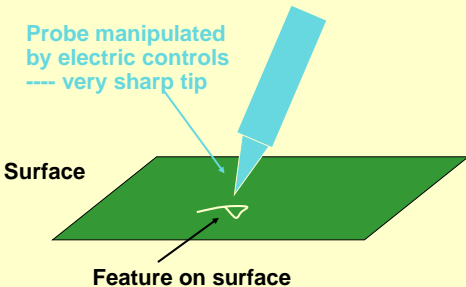
Liquid Helium at very low temperatures

Nobel Prize 2003 !
 Tony Leggett of this Physics Department

“Seeing” Quantum Effects in Our World

“Scanning Tunneling Microscope”
 Measures electric current from tip to surface as tip is moved

Probe manipulated by electric controls
 ---- very sharp tip



Surface

Feature on surface

“Seeing” Quantum Effects in Our World

Scanning Tunneling Microscope -- Nobel Prize 1985

Tip

Single atom at tip

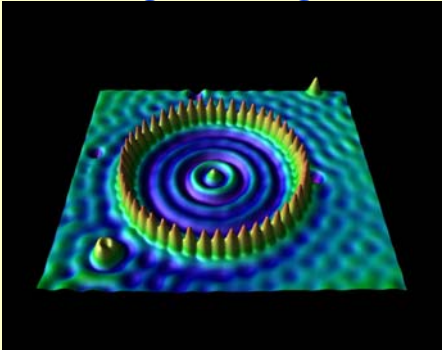
Electrons “Tunnel” from tip to surface

Extra atom on surface

Surface

Rate of tunneling extremely sensitive to distance of tip from surface due to quantum effects

Observation of atoms, electron waves with Scanning Tunneling Microscope



Observation of atoms, electron waves with Scanning Tunneling Microscope

Corral of atoms placed one at the time by maneuvering atoms with STM

Electron standing waves inside the “corral”

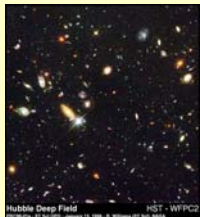
Extra atom

Surface Atoms

Figure by D. Eigler and coworkers, IBM Research

Lect. 19 - Big Picture: Smallest objects to the Universe

How do we know
the universe is expanding?
What galaxies are made of?



Distant galaxies seen
by Hubble Telescope



Exploding White Dwarf

Our understanding depends on
relativity and quantum mechanics!

Summary

- Enormous Ranges of sizes of objects in our world
- Enormous range of forces
- **Quantum Mechanics** crucial to understand the small units from which matter is made
 - Crucial for understanding Lasers, Semiconductors, Superconductors,
 - Atoms, Electrons, Nuclei, (More next time)
 - Quantum Effects NOT discernable for motions of ordinary objects (people, baseballs, sugar cubes,
- **Relativistic Effects** are important for objects moving very fast or very massive (more later)
 - speeds near $c = 10^8$ m/s
 - Protons, Neutrons inside nuclei have very high energies
 - Important for motions of galaxies at large fractions of c
 - Strong effects of gravity -- black holes