Instructors:
Eugene V Colla
kolla@illinois.edu
Anne Sickles
sickles@illinois.edu

Brian J Enders
benders2@illinois.edu
Abid Khan
aakhan3@illinois.edu
Hassan, Albur
ahassan4@illinois.edu

Shubhang
Goswami Shubhang
sqoswam3@illinois.edu

Chad Lantz
clantz2@illinois.edu

Laboratory Specialist:
Jack Boparai
jboparai@illinois.edu

Cristopher Zeitler
czeitle2@illinois.edu
Michelle Victora
victora2@illinois.edu
Joseph Chapman
jchapmn2@illinois.edu

Andrew Conrad
aconrad5@illinois.edu
Spencer Johnson
sij3@illinois.edu
I. **Goals of the course**

II. Teamwork / grades / expectations from you

III. Syllabus and schedule

IV. Your working mode
   - In class and “after hours” access
   - Safety, Responsibility
   - Home and away computing

V. Take a Lab tour!

VI. Let’s get started
   - electronic logbooks
   - digital scopes
Course Goals. Primary goals:

- **Learn how to “do” research**
  - Each project is a mini-research effort
  - How are experiments actually carried out?
    - The procedures aren’t all written out
    - The questions are not in the back of the chapter
    - The answers are not in the back of the book
    - You will have to learn to guide your own activities
  - Use of modern tools and modern analysis and data-recording techniques
Course Goals. Primary goals:

• Learn how to document your work
  • Online - electronic logbook *
  • Online – saving data and projects in student area on server
  • Using traditional paper logbooks
  • Making an analysis report
  • Writing formal reports
  • Presenting your findings orally

* In red gradable assignments
Course Goals. Secondary goals:

• **Learn some modern physics**
  
  – Many experiments were once Nobel-prize-worthy efforts
  
  – They touch on important themes in the development of modern physics
  
  – Some will provide additional insight to understand advanced courses you have taken
  
  – Some are just too new to be discussed in textbooks
The Experiments. Three main groups

- **Nuclear / Particle (NP)**
- **Atomic / Molecular / Optics (AMO)**
- **Condensed Matter (CM)**

You will do the experiment from all these groups
The Experiments

- **Nuclear / Particle (NP)**
  - Alpha particle range in gasses
  - $\gamma - \gamma$ correlation experiment
  - $\gamma$ – spectroscopy
  - Mössbauer spectroscopy
The Experiments

• **Nuclear / Particle (NP)**
  
  – Cosmic ray muons:
    
    Lifetime, capture rate, magnetic moment
  
  – Angular distribution of cosmic rays
  
  – $\gamma$ – spectroscopy
  
  – Mössbauer spectroscopy (new)
The Experiments

- **Condensed Matter (CM)**
  - Superconductivity
  - Tunneling in superconductors
  - 2\textsuperscript{nd} sound in 4He superfluid state
The Experiments

- **Condensed Matter (CM)**
  - Ferroelectrics and ferroelectric phase transition
  - Pulsed NMR
  - Calibration of temperature sensors

![Graph showing domains in BaTiO3 doped by Rh](image1.png)

![Graph showing P vs. E for BaTiO3](image2.png)
The Experiments

- **Condensed Matter (CM)**
  - Special Tools:
  - Vacuum film deposition
  - Atomic Force Microscope
  - Polarizing microscope
The Experiments
Atomic/Molecular/Optics (AMO)

- Berry’s phase
- Quantum erasure
- Quantum Entanglement
The Experiments

Atomic/Molecular/Optics (AMO)

- Optical pumping of rubidium gas
- Fluorescence spectroscopy
The “manuals”

• Many are just guides
• A only few purchased experiments have “real” manuals
• We serve as your guides … like real research

OPTICAL PUMPING OF RUBIDIUM OP1-A
Outline

I. Goals of the course
II. Teamwork / grades / expectations from you
III. Syllabus and schedule
IV. Your working mode
   In class and “after hours” access
   Safety, Responsibility
   Home and away computing
V. Take a Lab tour!
VI. Let’s get started
   electronic logbooks
   digital scopes
## Grading: Distribution of “1000” points

<table>
<thead>
<tr>
<th>ASSIGNMENT</th>
<th>Points</th>
</tr>
</thead>
<tbody>
<tr>
<td><strong>Expt. documentation</strong>: eelog reports, shift summaries, plot quality; paper logbooks</td>
<td>180 Total</td>
</tr>
<tr>
<td><strong>Formal reports</strong>: physics case, quality of results, depth of analysis, conclusions</td>
<td>600 Total</td>
</tr>
<tr>
<td><strong>Oral reports</strong>: motivation, organization of presentation; fielding questions</td>
<td>100 Total</td>
</tr>
<tr>
<td><strong>Final Oral Presentation = Final Exam</strong></td>
<td>120</td>
</tr>
<tr>
<td><strong>Total</strong></td>
<td>1000</td>
</tr>
<tr>
<td><strong>Effective point total will be</strong></td>
<td>1000</td>
</tr>
</tbody>
</table>

The grading scale will be a percentage out of “1000”:

Letter grading scale is approximately

- 97% = A+, 93% = A, 90% = A-, 87% = B+, 83% = B, 80% = B-, etc

You can **RESUBMIT one lab report** to improve your grade

(deadline for resubmissions May 5th)
Grading: a piece of history and analysis of the results

Physics 403
2005-2018
479 students total

Count

final grade

Physics 403
2005-2018
479 students total
Submission of Lab-Reports

• Due dates as on syllabus at midnight

• The reports should be uploaded to the server:
  • [https://my.physics.illinois.edu/courses/upload/](https://my.physics.illinois.edu/courses/upload/)

• Accepted MS-Word or PDF

• For orals – MS-PowerPoint or PDF
Absences

• If you are sick, let Eugene know by email (kolla@Illinois.edu). Don’t come in and get others sick. We are working side-by-side in a close environment for many hours.

• You can “make up” the time with arrangements and you can have access to the rooms. We will be accommodating.
Absences. Excuse Policy.

- You can be excused from only one missed assignment, and only if you provide medical documentation.
- If the excused assignment is an oral presentation, you will receive half the credit earned by your partner (this credit is for the effort you made in preparing the presentation file).
- The Final Oral cannot be excused, as it is equivalent to a final exam. You cannot pass the course without credit for this assignment (see Student Code).
Late Reports

• Policy for late reports

  ➢ You can have ONE “late ticket” for a “free” delay of up to 3 business days, but you must tell us you are using the ticket

  ➢ Reports are due at midnight on the date shown on the syllabus. After that we will charge:

    • 5 points for up to 1 week late. 10 points for up to 2 weeks late.

    • After that,
Outline

I. Goals of the course
II. Teamwork / grades / expectations from you
III. Syllabus and schedule
IV. Your working mode
   In class and “after hours” access
   Safety, Responsibility
   Home and away computing
V. Take a Lab tour!
VI. Let’s get started
    electronic logbooks
digital scopes
# Syllabus

## Cycles

Lecture topics are subject to change.

<table>
<thead>
<tr>
<th>Date</th>
<th>Day</th>
<th>Activity</th>
<th>Comment</th>
<th>Note</th>
</tr>
</thead>
<tbody>
<tr>
<td>1/15</td>
<td>Tues</td>
<td>Orientation</td>
<td>About Phy403 (ec)</td>
<td></td>
</tr>
<tr>
<td>1/17</td>
<td>Thurs</td>
<td>Cycle 1-1</td>
<td></td>
<td></td>
</tr>
<tr>
<td>1/22</td>
<td>Tues</td>
<td>Cycle 1-2</td>
<td>OriginPro Intro/Root (ec/as)</td>
<td></td>
</tr>
<tr>
<td>1/24</td>
<td>Thurs</td>
<td>Cycle 1-3</td>
<td>Elog Comments (ec/as)</td>
<td></td>
</tr>
<tr>
<td>1/29</td>
<td>Tues</td>
<td>Cycle 1-4</td>
<td>Written Reports (ec)</td>
<td></td>
</tr>
<tr>
<td>1/31</td>
<td>Thurs</td>
<td>Cycle 1-5</td>
<td></td>
<td>Rotate</td>
</tr>
<tr>
<td>2/5</td>
<td>Tues</td>
<td>Cycle 1-6</td>
<td>Error analysis (as)</td>
<td>C1-Ex1(2.06.19)</td>
</tr>
<tr>
<td>2/7</td>
<td>Thurs</td>
<td>Cycle 1-7</td>
<td></td>
<td></td>
</tr>
<tr>
<td>2/12</td>
<td>Tues</td>
<td>Cycle 1-8</td>
<td>Oral Reports/Talks(ec/as)</td>
<td></td>
</tr>
<tr>
<td>2/14</td>
<td>Thurs</td>
<td>Cycle 2-1</td>
<td></td>
<td>Rotate</td>
</tr>
<tr>
<td>2/19</td>
<td>Tues</td>
<td>ORALS Cycle 1</td>
<td></td>
<td></td>
</tr>
<tr>
<td>2/21</td>
<td>Thurs</td>
<td>Cycle 2-2</td>
<td></td>
<td>C1-Ex2 (2.27.19)</td>
</tr>
<tr>
<td>2/26</td>
<td>Tues</td>
<td>Cycle 2-3</td>
<td>Optical spectroscopy</td>
<td></td>
</tr>
<tr>
<td>2/28</td>
<td>Thurs</td>
<td>Cycle 2-4</td>
<td></td>
<td></td>
</tr>
<tr>
<td>3/5</td>
<td>Tues</td>
<td>Cycle 2-5</td>
<td>Measuring Temp (ec)</td>
<td>Rotate</td>
</tr>
<tr>
<td>3/7</td>
<td>Thurs</td>
<td>Cycle 2-6</td>
<td></td>
<td></td>
</tr>
<tr>
<td>3/12</td>
<td>Tues</td>
<td>Cycle 2-7</td>
<td>Entanglement</td>
<td>C2-Ex1 (3.13.19)</td>
</tr>
<tr>
<td>3/14</td>
<td>Thurs</td>
<td>Cycle 2-8</td>
<td></td>
<td></td>
</tr>
<tr>
<td>3/16</td>
<td></td>
<td>Spring Break</td>
<td></td>
<td></td>
</tr>
<tr>
<td>3/26</td>
<td>Tues</td>
<td>Cycle 3-1</td>
<td>Noise (mw)</td>
<td>Rotate</td>
</tr>
<tr>
<td>3/28</td>
<td>Tues</td>
<td>Cycle 3-2</td>
<td></td>
<td></td>
</tr>
<tr>
<td>4/2</td>
<td>Tues</td>
<td>ORALS Cycle 2</td>
<td></td>
<td>C2-Ex2 (4.5.19)</td>
</tr>
<tr>
<td>4/4</td>
<td>Thurs</td>
<td>Cycle 3-3</td>
<td></td>
<td></td>
</tr>
<tr>
<td>4/9</td>
<td>Tues</td>
<td>Cycle 3-4</td>
<td>Lock-in Amps and FT(ec)</td>
<td></td>
</tr>
<tr>
<td>4/11</td>
<td>Thurs</td>
<td>Cycle 3-5</td>
<td></td>
<td>Rotate</td>
</tr>
<tr>
<td>4/16</td>
<td>Tues</td>
<td>Cycle 3-6</td>
<td>Ferroelectricity (ec)</td>
<td>C3-Ex1 (4.17.19)</td>
</tr>
<tr>
<td>4/18</td>
<td>Thurs</td>
<td>Cycle 3-7</td>
<td></td>
<td></td>
</tr>
<tr>
<td>4/23</td>
<td>Tues</td>
<td>Cycle 3-8</td>
<td>To be announced</td>
<td></td>
</tr>
<tr>
<td>4/25</td>
<td>Thurs</td>
<td>Final Orals #1</td>
<td></td>
<td></td>
</tr>
<tr>
<td>4/30</td>
<td>Tues</td>
<td>Final Orals #1</td>
<td></td>
<td></td>
</tr>
<tr>
<td>5/2</td>
<td></td>
<td>READING DAY</td>
<td></td>
<td>C3-Ex2 (5.3.19)</td>
</tr>
</tbody>
</table>

* Physics 403*
<table>
<thead>
<tr>
<th>NP</th>
<th>CM</th>
<th>Atomic + CM</th>
<th>Optics</th>
</tr>
</thead>
</table>
| A. Cosmic Muon Stand  
   i. Muon lifetime  
   ii. Capture rate  
   iii. Magnetic moment  
B. Alpha range  
C. Gamma-Gamma  
D. Cosmic angular distribution | A. Ferro 1  
B. Ferro 2 (imaging)  
C. 2nd sound of 4He  
D. pNMR  
E. Hysteresis loops  
F. Tunneling  
G. AFM  
H. T calibration | A. Optical pumping  
B. Superconductivity  
C. Mutual inductance  
D. Fluorescence spectroscopy | A. Quantum erasure  
B. Berry’s phase  
C. Entanglement |

<table>
<thead>
<tr>
<th>Anne, Chad</th>
<th>Eugene, Albur</th>
<th>Eugene, Brian, Abid, Shubhang</th>
<th>TA’s from Kwiat Lab</th>
</tr>
</thead>
<tbody>
<tr>
<td>C1-1 1- 2; 3- 4; 5- 6; 7- 8</td>
<td>9- 10; 11- 12; 13- 14; 15- 16</td>
<td>17- 18; 19- 20; 23- 24</td>
<td>21- 22; 25- 26</td>
</tr>
<tr>
<td>C1-2 1- 2; 3- 4; 5- 6; 7- 8</td>
<td>9- 10; 11- 12; 13- 14; 15- 16</td>
<td>17- 18; 21- 22; 25- 26</td>
<td>19- 20; 23- 24</td>
</tr>
<tr>
<td>C2-1 9- 12; 11- 14; 13- 16; 15- 10</td>
<td>17- 20; 19- 22; 21- 24; 23- 26; 25- 18</td>
<td>3- 6; 7- 2</td>
<td>1- 4; 5- 8</td>
</tr>
<tr>
<td>C2-2 9- 12; 11- 14; 13- 16; 15- 10</td>
<td>17- 20; 19- 22; 21- 24; 23- 26; 25- 18</td>
<td>1- 4; 5- 8</td>
<td>3- 6; 7- 2</td>
</tr>
<tr>
<td>C3-1 17- 22; 19- 24; 21- 26; 23- 18; 25- 20</td>
<td>1- 6; 3- 8; 5- 2; 7- 4</td>
<td>9- 14; 13- 10</td>
<td>11- 16; 15- 12</td>
</tr>
<tr>
<td>C3-2 17- 22; 19- 24; 21- 26; 23- 18; 25- 20</td>
<td>1- 6; 3- 8; 5- 2; 7- 4</td>
<td>11- 16; 15- 12</td>
<td>9- 14; 13- 10</td>
</tr>
</tbody>
</table>
Assignment of experiments

3 cycles with 2 experiments

→ teams change after cycle

→ **joint team** reports and oral presentations

→ Final Oral presentation will be done by each student **personally**
After 2 experiments (1 cycle) we will have oral session. The topic of the presentation will be chosen from the experiments done in this cycle.

<table>
<thead>
<tr>
<th>Cycle</th>
<th>#</th>
<th>Experiment</th>
</tr>
</thead>
<tbody>
<tr>
<td>C1-1</td>
<td>1,2</td>
<td>Cosmic Muon</td>
</tr>
<tr>
<td>C1-2</td>
<td>1,2</td>
<td>Gamma-Gamma</td>
</tr>
</tbody>
</table>

**Fluorescence of Ruby Crystals**

*Student#1 and Student #2*
Outline

I. Goals of the course
II. Teamwork / grades / expectations from you
III. Syllabus and schedule
IV. Your working mode
   In class and “after hours” access
   Safety, Responsibility
   Home and away computing
V. Take a Lab tour!
VI. Let’s get started
   electronic logbooks
   digital scopes
Lab Access

Use Your ID Card to Access the Lab

You can access the Lab not only on “Lab days”

Late time rules:

You can stay in the Lab until 8pm but need to work with partner

After 8pm and on weekend days – you have to discuss this schedule with your instructor and in general it is preferable to avoid working after 8 pm and on week
Safety is your responsibility!

Hazards: *high voltage, radioactive sources, cryogens, chemical materials, high pressure*

In class work and “after hours” access & work requires responsible conduct with regards to

(I) safety/hazards and with

(II) equipment

Discuss potential hazards at the beginning of each experiment with an instructor or TA

When in doubt stop and ask

Problems after hours: 217 493 1576 (Eugene’s cell)

217 979 4990 (Anna’s cell)
Follow Directly the Recommendations of Safety Working

https://www.drs.illinois.edu/
Follow Directly the Recommendations of Safety Working

Chemical Waste Collection and Storage

Before generating chemical waste, the researcher should determine how it will be collected and stored and obtain the necessary equipment (containers, labels) in advance. The choice of procedures depends on the type of waste and its final disposition. This section explains how to determine the final disposition of waste, select the appropriate waste container, and store waste in the lab or work area. It also suggests waste minimization strategies.

Determining How to Dispose of a Chemical Waste

The final disposition of a chemical waste is determined by the answers to a series of questions:

Step 1. Is the waste Contaminated Debris (glassware, paper towels, clean-up materials), or is it a chemical or chemical mixture?
   If it is contaminated debris: Go to Step 5.
   If it is a chemical or chemical mixture: Go to Step 2.

Step 2. Is the chemical a DEA (Drug Enforcement Agency) controlled substance? (Refer to the DEA list controlled substances [ ]) Yes: Refer to the DEA Controlled Substances Guide for disposal procedures.
   No: Go to Step 3.

Step 3. Is the chemical a solid (not liquid or gas)?
   Yes: Collect and store the waste as described in the waste container and storage guidelines listed below, and dispose of it through the Division of Research Safety (DRS) chemical waste disposal program. See the section Procedures for Requesting Chemical Waste Disposal for the disposal procedures. (No solid chemical waste, hazardous or non-hazardous, should be placed in the regular trash.)
   No: Go to Step 4.

Step 4. Is the chemical a liquid non-hazardous waste as listed in the section Liquid Non-Hazardous Chemical Waste Disposal?
   Yes: The chemical may be poured down the sanitary sewer (sink drain) with copious amounts of water.
   No: Collect and store the waste as described in the waste container and storage guidelines listed below, and dispose of it through the DRS chemical waste disposal program. See the section Procedures for Requesting Chemical Waste Disposal for the disposal procedures.

Step 5. Is the contaminated debris laboratory glassware (broken and unbroken)?
   Yes: See the Laboratory Glassware Waste Disposal section.
   No: Go to Step 6.

Step 6. Is the debris contaminated with a substance listed in the section Liquid Non-Hazardous Chemical Waste Disposal?
   Yes: The contaminated debris can be disposed of in the regular frozen storage.
   No: Collect and store the contaminated debris as described in the waste container and storage guidelines listed below, dispose of it through the DRS chemical waste disposal program.
Definition

Materials that qualify as “sharps” are defined at the state level and shall be disposed of as Potentially Infectious Medical Waste (PIMW). In Illinois, the Illinois Environmental Protection Agency (IEPA) has designated the following material (used or unused) as sharps:

- Any medical needles,
- Syringe barrels (with or without needle),
- Pasteur pipettes (glass),
- Scalpel and razor blades,
- Blood vials,
- Microscope slides and coverslips,
- Glassware contaminated with infectious agents.

NEVER dispose of these items in SDCs:

- Plastic items (except for syringes),
- Beverage containers (no pop cans!),
- Non-biologically contaminated laboratory glassware,
- Solvent/chemical bottles,
- Light bulbs,
- Any paper materials,
- Pipette tips,
- Plastic pipettes,
- Aerosol cans or cans of any type,
- Scintillation vials,
- Any item with liquid (except for blood in vacutainer tubes).

Waste container for sharps
Outline

V. Take a Lab tour!

VI. Let’s get started
electronic logbooks
digital scopes
Outline

I. Goals of the course
II. Teamwork / grades / expectations from you
III. Syllabus and schedule
IV. Your working mode
   In class and “after hours” access
   Safety, Responsibility
   Home and away computing
V. Take a Lab tour!
VI. Let’s get started
   electronic logbooks
digital scopes
How to record the data

• Work together

• Write down the equipment used

• Make a diagram of the setup

• Note the settings of dials, switches, gauges

• Take a digital photo if appropriate

• Use a software drawing program to make a detailed sketch
How to record the data

- Use the eLog (see next).
- Write down what you did in real sentences.
- Provide enough detail that you can reconstruct later what you did!
- How will you look at the data later?
- Do you have enough information?
- Did the equipment perform as expected?
How to record the data

• Many experiments require you to “change and measure” something by hand
  – Make a **table** in a **paper logbook** for this
  – Be prepared to state your measurement uncertainty
  – Make a “**quick sketch**” of your results by hand; then, enter the data in an electronic table and make a final plot
  - Do you have enough points?
  - Do you have any obvious anomalies?
  - You can repeat points but do not throw them out. Use other measurements to check reliability
Many experiments have built-in, computer-based data acquisition (DAQ)

- You will not have time to fully understand the DAQ, but

  • Be sure you know functionally what it is doing – ask
  • A good idea is to make test measurements of something you know
    - As before, anomalies? enough points? uncertainties?
Where to exchange, store and retrieve course information.

(i) Your data, projects, tables etc

\`\textbackslash engr-file-03\textbackslash PHYINST\textbackslash APL Courses\textbackslash PHYCS403\`

Each student has a folder

Store all experiment related materials in corresponding folder
Where to exchange, store and retrieve course information. *(i)*

Your data, projects, tables etc

An example of the “smart” structure of folders containing the raw data and data analysis projects.
Where to retrieve course information.

**Manuals, papers, setup diagrams and other useful materials**
Where to retrieve course information.

Manuals, papers, **setup diagrams** and other useful materials

α-range experiment setup

α-range experiment setup diagram
Where to retrieve course information.

Setup diagrams – do not use cellphones to take the image of the setup from manual – for most setups we have PowerPoint projects with setups.
Where to retrieve course information.

Manuals, papers, setup diagrams and other useful materials

- Some old stuff (not very useful)
- Sample pictures of ferroelectric domains
- Examples of report and oral presentation
- Pictures of the setups of the experiments
- Software including DAQ software for different experiments. Newest version of Origin is also there
- P403 lecture notes
- C++ scripts for Root
- Origin manuals + a very compressed version written by Eugene
- Origin templates (how to use them will be discussed in next lecture)
Connecting to the PHYS403 server

Connect to VPN following the instructions on the UIUC VPN website:
https://techservices.illinois.edu/services/virtual-private-networking-vpn/download-and-set-up-the-vpn-client

To connect to the PHYS403 Server:

• Connect to the VPN first, then enter the following as the share to connect to:
  – **Mac users:** Open Finder: Go: Connect to Server, type in address:
    smb://engr-file-03.engr.illinois.edu/PHYINST/APL Courses/PHYCS403
  – **Windows users:** Open Windows Explorer, type in address:
    \engr-file-03.engr.illinois.edu\PHYINST\APL Courses\PHYCS403
    or, if that doesn’t work:
    //engr-file-03.engr.Illinois.edu/PHYINST/APL Courses/PHYCS403

• **When prompted for username and password, enter:**
  “UofI\[your netID]” and “[your netID password]”
“Journal club”

http://ajp.aapt.org/#mainWithRight

http://www.nature.com/nature/index.htm

http://www.scientificamerican.com/

http://www.sciencemag.org/journals

http://publish.aps.org  or  http://prola.aps.org/
“Journal club”

Walking with Coffee: Why Does it Spill?

Growth of Diamond Films from Tequila

J. Morales\textsuperscript{1,2}, L. M. Apátiga\textsuperscript{2}, V. M. Castaño\textsuperscript{2}

1. Facultad de Ciencias Físico Matemáticas, Universidad Autónoma de Nuevo León
2. Centro de física Aplicada y Tecnología Avanzada, Universidad Nacional Autónoma de México

Fabrication and Characterization of Ultrathin Three-Dimensional Thermal Cloak

Student #1
University of Illinois at Urbana-Champaign

The Physics of Beer Tapping

PRESENTATION BY JOSEPH MIRABELLI
JAVIER RODRÍGUEZ-RODRÍGUEZ, J.*, ALMUDENA CASADO-CHACÓN, AND DANIEL PUSTER
1. FLUID MECHANICS GROUP, CARLOS III UNIVERSITY OF MADRID
2. CNRS, UNIVERSITÉ PIERRE ET MARIE CURIE
“Journal club”

**Journal Access**
If you cannot access journal papers using VPN, go to UIUC’s library proxy test site and enter the address of the paper you want to read:
http://www.library.illinois.edu/proxy/test/

**Recommended journal websites**

- American Physical Society Journals: [https://journals.aps.org/about](https://journals.aps.org/about)
- Nature: [http://www.nature.com/nature/index.html](http://www.nature.com/nature/index.html)
- Science: [http://www.sciencemag.org/journals](http://www.sciencemag.org/journals)
- American Journal of Physics: [http://scitation.aip.org/content/aapt/journal/ajp](http://scitation.aip.org/content/aapt/journal/ajp)
Entering the e-Log ...

PHYS 403 Spring 2019

Welcome
Please see the course description for an explanation of how this course works. It may seem first, but all the pieces do work together to enhance understanding. Also, please consult the help you keep track of what is due when. Please take a moment to review the welcome pa should give you a good overview of what to expect to learn in this course.
Entering the e-Log ...

Use your University Username and Password
## Modern Physics Laboratory Logbook

<table>
<thead>
<tr>
<th>Logbook</th>
<th>Entries</th>
<th>Last submission</th>
</tr>
</thead>
<tbody>
<tr>
<td><strong>phys497EVC-2017</strong></td>
<td></td>
<td></td>
</tr>
<tr>
<td>PHYS 497 EVC Spring 2017</td>
<td>1</td>
<td>02/23/17 17:00 by Matthew Podkowa, Tianqi Gu</td>
</tr>
<tr>
<td>PHYS 497 EVC Summer 2017</td>
<td>0</td>
<td>-</td>
</tr>
<tr>
<td>PHYS 497 EVC Fall 2017</td>
<td>4</td>
<td>11/16/17 14:29 by Oliver Wang, Patrick Bullock</td>
</tr>
<tr>
<td><strong>phys403-2019</strong></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Physics 403_Spring 2019</td>
<td>0</td>
<td>-</td>
</tr>
<tr>
<td><strong>phys403-2018</strong></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Physics 403_Spring 2018</td>
<td>307</td>
<td>11/01/18 14:00 by Joseph Bryant, Aike Liu</td>
</tr>
</tbody>
</table>
# Entering the e-Log ...

**Message ID:** 1    **Entry time:** 01/14/19 21:30

<table>
<thead>
<tr>
<th>Author</th>
<th>Eugene Colla</th>
</tr>
</thead>
<tbody>
<tr>
<td>Experiment</td>
<td>none</td>
</tr>
<tr>
<td>Post Type</td>
<td>Other</td>
</tr>
<tr>
<td>Subject</td>
<td>Welcome</td>
</tr>
</tbody>
</table>

Welcome to Physics 403 course
e-logs: First a brief tour

How to use it

• Pause and summarize your work at natural stopping points in the action. This is useful for particular findings and measurement sequences.

• Along the way, save data, plots, scope shots to your folder on the server.

• Near the end of the class, add a summary/conclusion, indicate future directions, and make sure the e-log provides a rather complete overview of the highlights of your work. Upload your plots, scope shots, etc. and describe the data.
e-logs: Making a post ...

- **Create a New Post**
- **To create a new post, click "New" from the menu bar.**
- **Fill in the **Author, Experiment, Post Type, and Subject**

If the post is written by more than one person, use a comma separated list.

Be sure the Author name is the same you used when registering so that you can edit/delete the post if necessary.
e-logs: Making a post ...

<table>
<thead>
<tr>
<th>Author:</th>
<th>Your name and your partner’s name</th>
</tr>
</thead>
<tbody>
<tr>
<td>Experiment:</td>
<td>General</td>
</tr>
<tr>
<td>Post Type:</td>
<td>How-To</td>
</tr>
<tr>
<td>Subject:</td>
<td>Day [#]: brief description of work</td>
</tr>
</tbody>
</table>

Goal: Be specific. Not, “Learn about experiment,” but, for example, “In helium below temperatures of 2.17K, a second sound due to thermal effects becomes measurable. We will measure second sound using a resonant cavity…”

Settings / Equipment Notes: Note important environmental and experimental parameters such as atmospheric pressure, settings on equipment, etc.

[Time Range 1]: Give time range, not just “before tea.”
- Note important steps and results
- Include plots, photos, or scope shots in attachments below
- Use bullet points to make it easy to read

[Time Range 2]: ...

Conclusions & Future Plans: What did you find and what is the next step? Be specific. Not, “We measured decay times,” but, for example, “Ruby #2 sample with higher concentration chromium was observed to decay with a form…”

Physics 403