

UNIVERSITY OF ILLINOIS
AT URBANA-CHAMPAIGN

Basic Error Analysis

Physics 403

Eugene V Colla

D. Hertzog

A. Bezryadin



illinois.edu

Outline of the lecture

- **Errors and uncertainties**
- **The reading error**
- **Accuracy and precession**
- **Systematic and statistical errors**
- **Fitting errors**
- **Presentation of the results**
- **Heisenberg limit precision measurements**



Introduction

- Uncertainties exist in all experiments
- The final goal of any experiment is to obtain *reproducible* results. Knowing errors and uncertainties is an essential part for ensuring reproducibility.
- To know the uncertainties we use two approaches:
 - (1) Repeat each measurement many times and determine how well the result reproduces itself. If the results are different then there are **statistical errors**.
 - (2) Measure the quantity of interest using a different method. The results, if correct, are independent of the measurement technique. If the results are different then there are **systematic errors** in one of the methods or in both.
- Presenting the result of your experiment: Use the right number of significant digits, in agreement with the estimated uncertainty.



Errors (uncertainties)

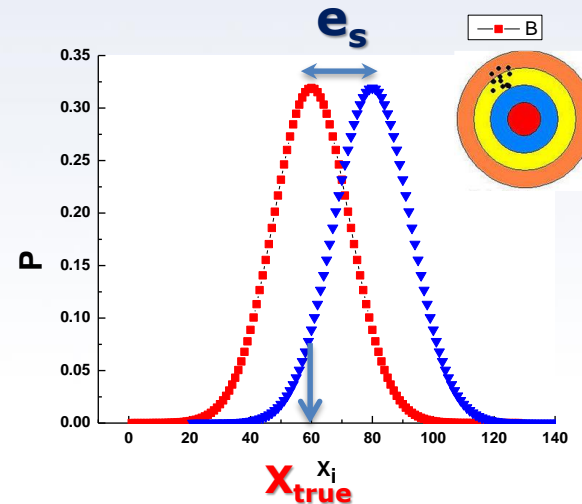
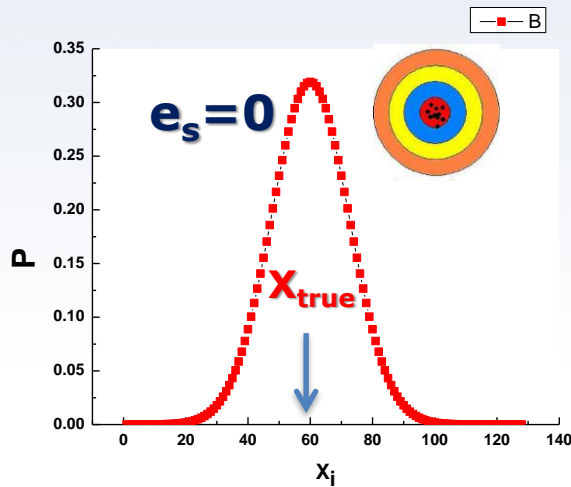
Result of measurement

$$X_{\text{meas}} = X_{\text{true}} + e_s + e_r$$

Correct value

Systematic error

Random error



Systematic vs. Statistical Uncertainties

- Systematic uncertainty
 - Uncertainties associated with imperfect knowledge of measurement **apparatus**, other physical **quantities** needed for the measurement, or the physical **model** used to interpret the data.
 - Generally correlated between measurements. Cannot be reduced by multiple measurements.
 - Better calibration, or measurements employing different techniques or methods can reduce the uncertainty.
- Statistical Uncertainty
 - Uncertainties due to stochastic fluctuations
 - Generally there is no correlation between successive measurements.
 - Multiple measurements can be used to reduce this uncertainty.



The Difference Between Systematic & Random Errors

- Random error describes errors that fluctuate due to the unpredictability or uncertainty inherent in your measuring process, or the variation in the quantity you're trying to measure. **Such errors can be reduced by repeating the measurement and averaging the results.**
- A systematic error is one that results from a persistent issue and leads to a consistent error in your measurements. For example, if your measuring tape has been stretched out, your results will always be lower than the true value. Similarly, if you're using scales that haven't been set to zero beforehand, there will be a systematic error resulting from the mistake in the calibration. **Such errors cannot be reduced simply by repeating the measurement and averaging the results. Such errors can be reduced by analyzing the instrument(s) used for the measurement and by using different instruments.**



Definitions (NIST)

The standard uncertainty σ of a measurement result x is the estimated standard deviation of x .

The relative standard uncertainty σ_r of a measurement result x is defined by $\sigma_r = \sigma / |x|$, where x is not equal to 0.

In statistics, the standard deviation (SD, also represented by the Greek letter sigma σ or the Latin letter s) is a measure that is used to quantify the amount of variation or dispersion of a set of data values. A low standard deviation indicates that the data points tend to be close to the mean value of the set ($\mu = \langle x_i \rangle$), while a high standard deviation indicates that the data points are spread out over a wider range of values.

$$\sigma = \sqrt{\frac{1}{N} \sum_{i=1}^N (x_i - \mu)^2}, \quad \text{where } \mu = \frac{1}{N} \sum_{i=1}^N x_i$$



Meaning

Meaning of uncertainty:

If the probability distribution characterized by the averaged measurement result \bar{y} and its standard uncertainty σ is approximately normal (Gaussian), and σ is the standard deviation of \mathbf{x} , then the interval $\mathbf{x} - \sigma$ to $\mathbf{x} + \sigma$ is expected to encompass approximately 68 % of the measurement results (data points).

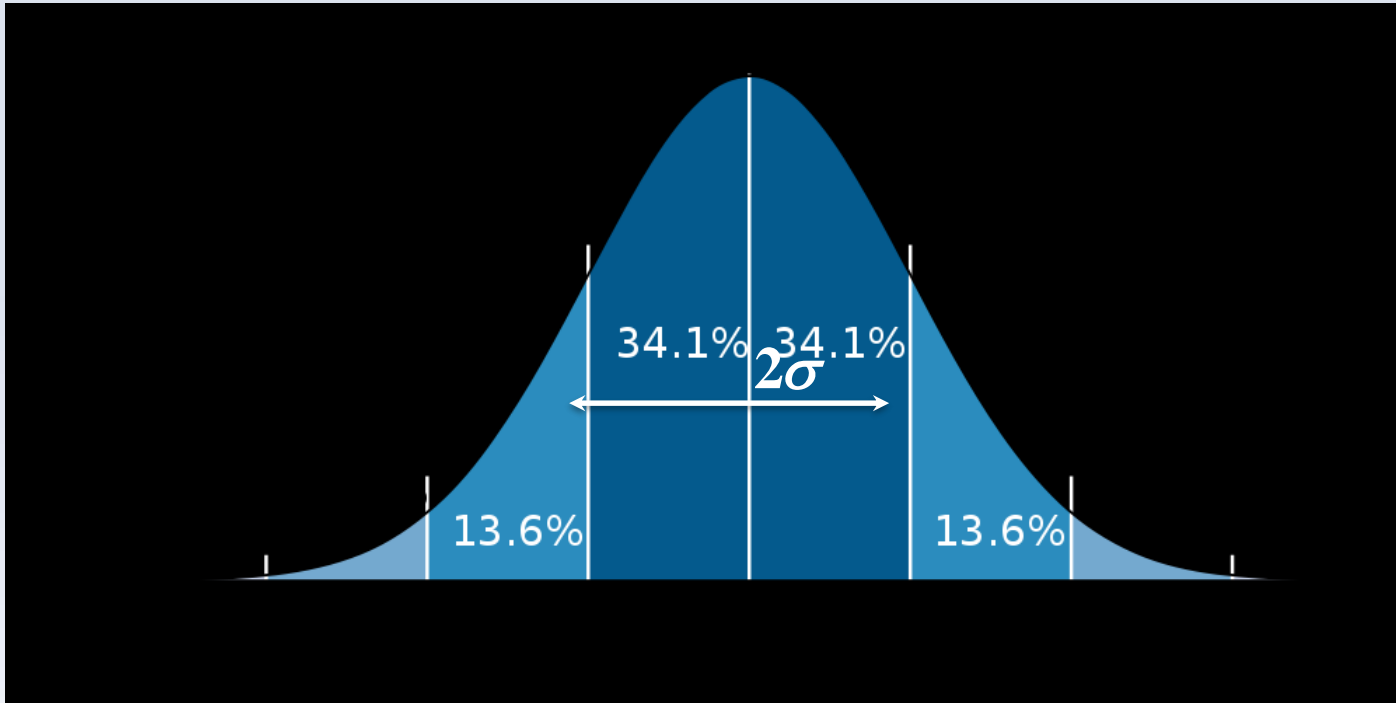
Here \mathbf{X} is the true value (never known exactly) and \mathbf{x} is the measured value.

The probability that the true value \mathbf{X} is greater than $\mathbf{x} - \sigma$, and is less than $\mathbf{x} + \sigma$ is estimated as 68%.

This statement is commonly written as $\mathbf{X} = \mathbf{x} \pm \sigma$.



Normal (Gaussian) distribution



$$P_n(x) = \frac{1}{\sigma\sqrt{2\pi}} e^{-\frac{(x-\bar{x})^2}{2\sigma^2}}$$

The interval representing two standard deviations contains 95.4% of all possible true values.

Confidence interval $\langle x \rangle \pm 3\sigma$ contains 99.7% of possible outcomes.



Notations

Use of concise notation:

If, for example, $v = 1\,234.567\,89\text{ m/s}$ and $\Delta v = 0.000\,11\text{ m/s}$, where m/s is the unit of v , then $v = (1\,234.567\,89 \pm 0.000\,11)\text{ m/s}$.

A more concise form of this expression, and one that is used sometimes, is $v = 1\,234.567\,89(11)\text{ m/s}$, where it is understood that the number in parentheses is the numerical value of the standard uncertainty referred to the corresponding last digits of the quoted result.

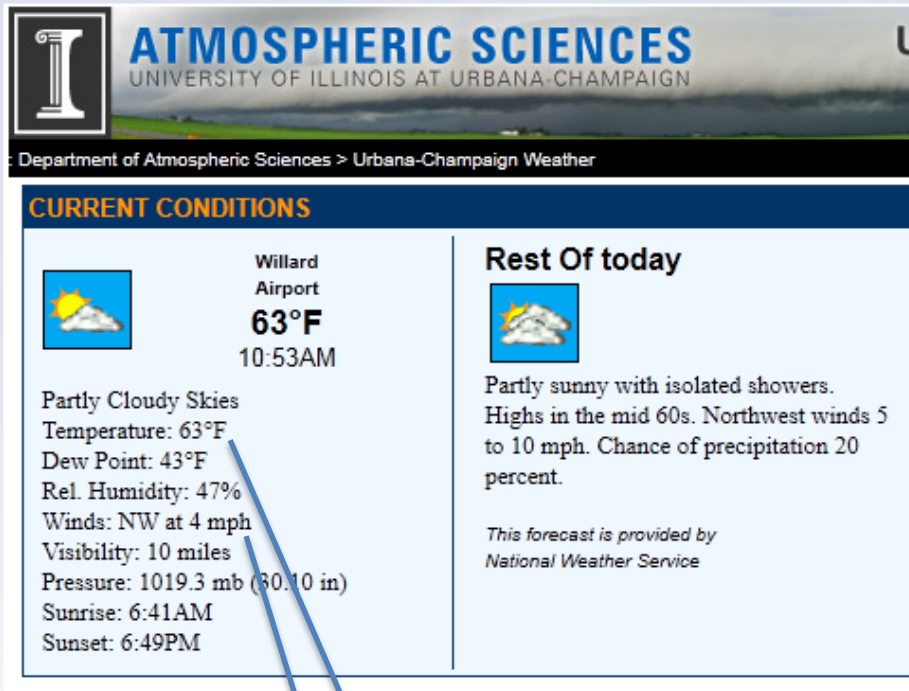
Examples of results which do not make sense (too many digits):

$$v = (1234.5678934534940945 \pm 0.011)\text{ m/s}$$

$$\text{or } v = (1234.56 \pm 2)\text{ m/s}$$





Significant digits



ATMOSPHERIC SCIENCES
UNIVERSITY OF ILLINOIS AT URBANA-CHAMPAIGN

Department of Atmospheric Sciences > Urbana-Champaign Weather

CURRENT CONDITIONS

 Willard Airport 63°F 10:53AM Partly Cloudy Skies Temperature: 63°F Dew Point: 43°F Rel. Humidity: 47% Winds: NW at 4 mph Visibility: 10 miles Pressure: 1019.3 mb (30.10 in) Sunrise: 6:41AM Sunset: 6:49PM	Rest Of today  Partly sunny with isolated showers. Highs in the mid 60s. Northwest winds 5 to 10 mph. Chance of precipitation 20 percent. <i>This forecast is provided by National Weather Service</i>
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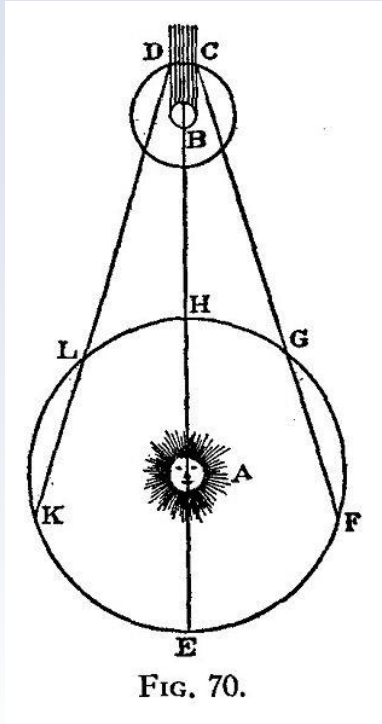
$T = 63^{\circ}\text{F} \pm ?$ \longrightarrow Best guess $\Delta T \sim 0.5^{\circ}\text{F}$

Wind speed $4\text{mph} \pm ?$ \longrightarrow Best guess $\pm 0.5\text{mph}$

If they say $T=63.32456\text{ F}$, that would be wrong since they cannot predict temperature with such high precision and the temperature is not stable up to so many significant digits



It is important to know uncertainties in science



Measurement of the speed of the light

1675 Ole Roemer: 220,000 Km/sec



Ole Christensen Rømer
1644-1710

Does it make sense?
What is missing?

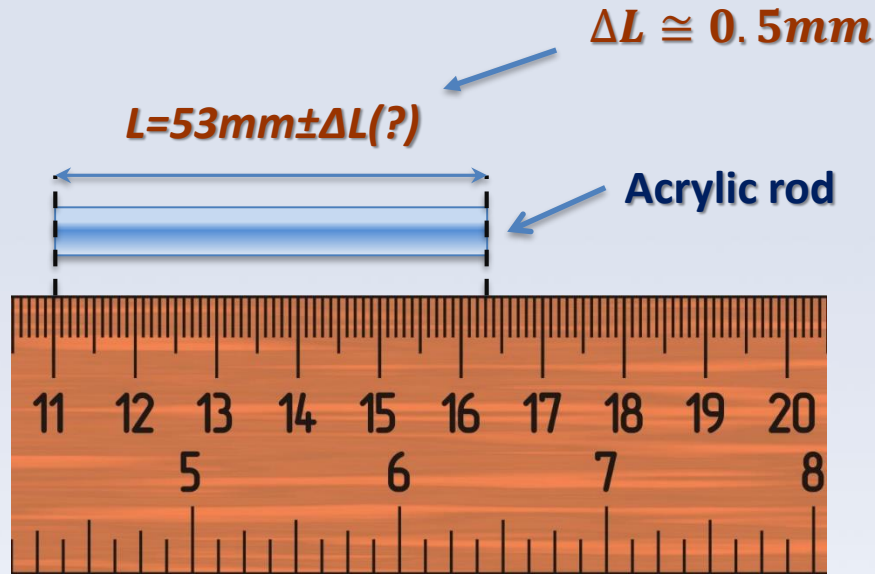
Maxwell's theory prediction:

Speed of light does not depend on the light wavelength. It is a universal constant.



NIST Boulder Colorado $c = 299,792,456.2 \pm 1.1$ m/s.

Reading error



$\Delta L \cong 0.03\text{mm}$



How far we have to go in reducing the reading error?

Use a simple ruler if you do not care about accuracy better than 1mm

Otherwise you need to use digital calipers

Probably the natural limit of accuracy can be due to length uncertainty because of temperature expansion. For 53mm $\Delta L \cong 0.012\text{mm}/K$



Reading Error = $\pm \frac{1}{2}$ (least count or minimum gradation).

Reading error. Digital meters.



Fluke 8845A multimeter

Example Vdc (reading)=0.85V

$$\Delta V = 0.85 \times (1.8 \times 10^{-5}) \sim 15\mu V$$

8846A Accuracy

Accuracy is given as \pm (% measurement + % of range)

Range	24 Hour (23 \pm 1 $^{\circ}$ C)	90 Days (23 \pm 5 $^{\circ}$ C)	1 Year (23 \pm 5 $^{\circ}$ C)	Temperature Coefficient/ $^{\circ}$ C Outside 18 to 28 $^{\circ}$ C
100 mV	0.0025 + 0.003	0.0025 + 0.0035	0.0037 + 0.0035	0.0005 + 0.0005
1 V	0.0018 + 0.0006	0.0018 + 0.0007	0.0025 + 0.0007	0.0005 + 0.0001
10 V	0.0013 + 0.0004	0.0018 + 0.0005	0.0024 + 0.0005	0.0005 + 0.0001
100 V	0.0018 + 0.0006	0.0027 + 0.0006	0.0038 + 0.0006	0.0005 + 0.0001
1000 V	0.0018 + 0.0006	0.0031 + 0.001	0.0041 + 0.001	0.0005 + 0.0001



Accuracy and precision



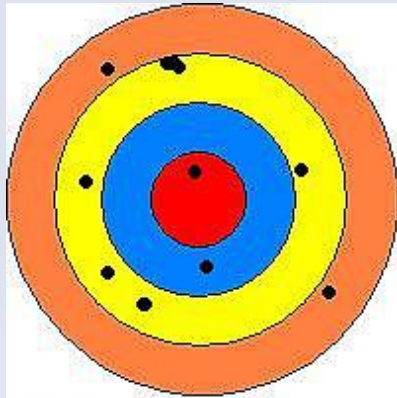
The accuracy of an experiment is a measure of how close the result of the experiment comes to the true value



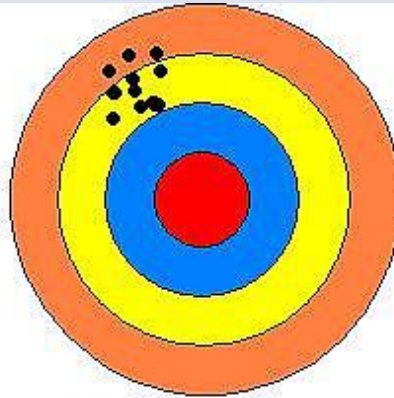
Precision refers to how closely individual measurements agree with each other



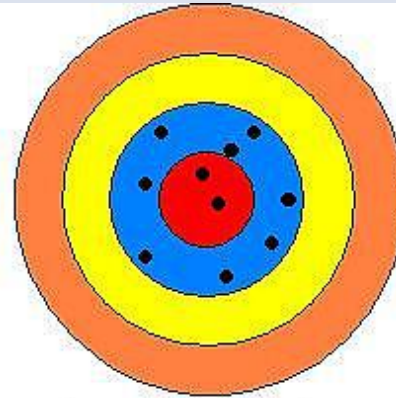
Accuracy and precision



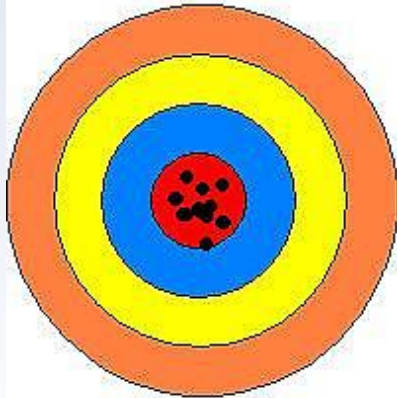
Not Precise, Not Accurate



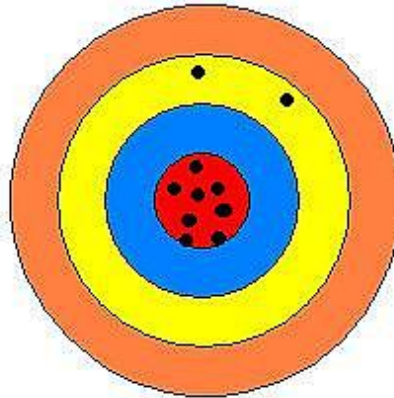
Precise, Not Accurate



Accurate, Not Precise



Accurate, Precise



Errors



Systematic and random errors

- ***Systematic Error***: reproducible inaccuracy introduced by faulty equipment, calibration, technique, model, drifts.
- ***Random errors***: Indefiniteness of results due to finite precision of experiment. Errors can be reduced by repeating the measurement and averaging. These errors can be caused by thermal motion of molecules and electrons in the apparatus.

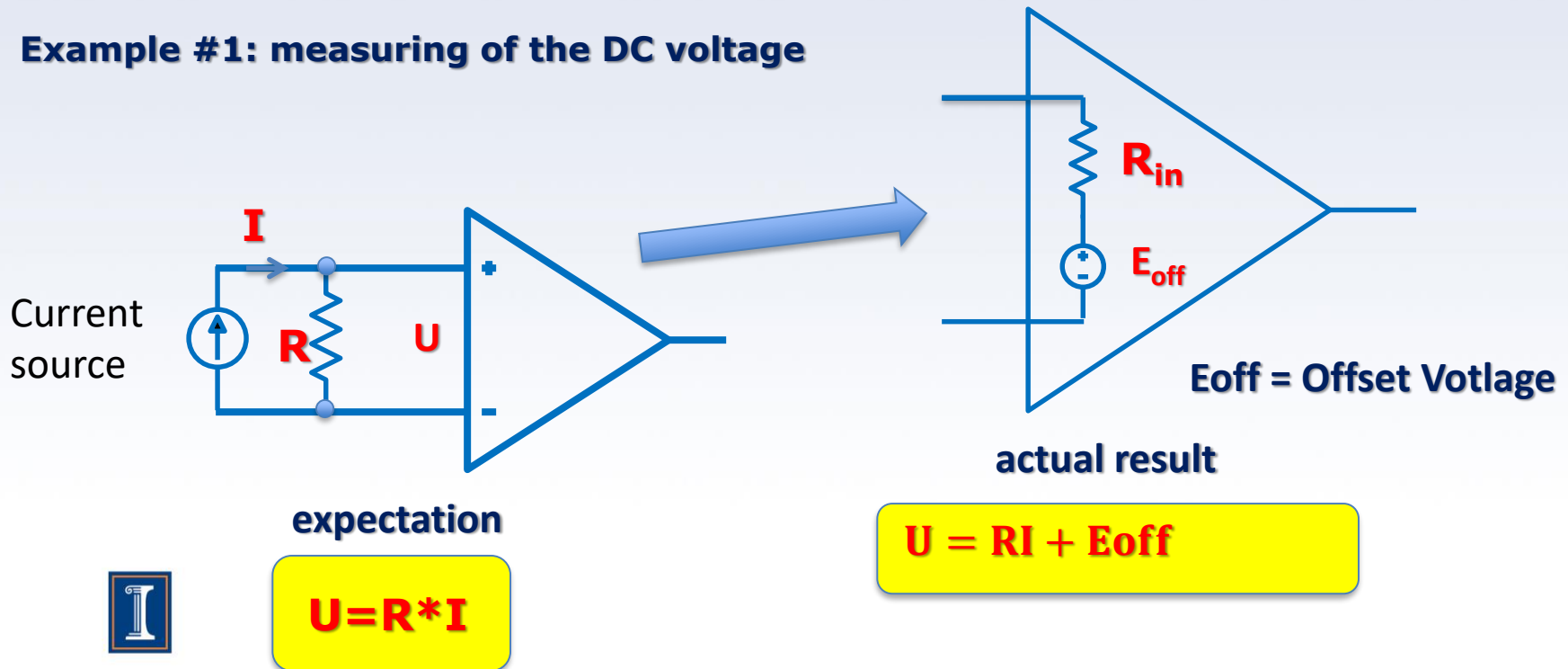
Philip R. Bevington “Data Reduction and Error Analysis for the Physical sciences”, McGraw-Hill, 1969



Systematic errors

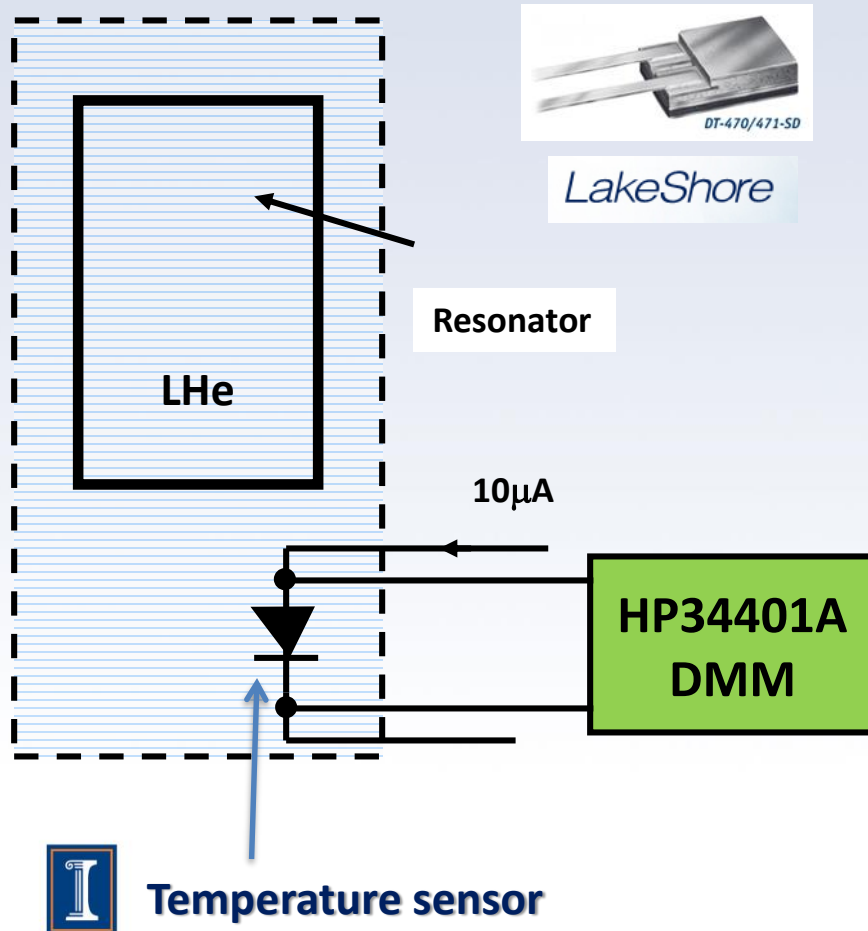
Sources of systematic errors: poor calibration of the equipment, changes of environmental conditions, imperfect method of observation, drift and some offset in readings etc.

Example #1: measuring of the DC voltage

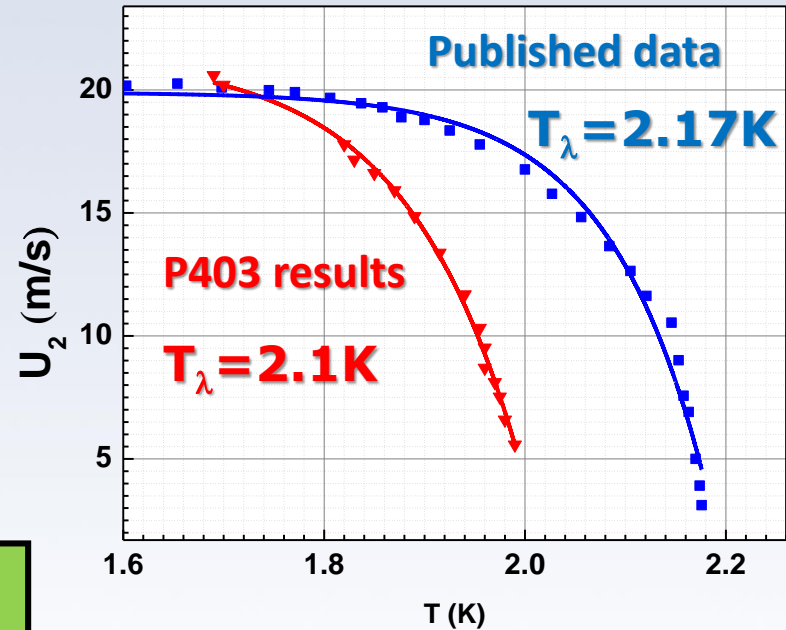


Systematic errors

Example #3: poor calibration



Measuring of the speed of the second sound in superfluid He4



Random errors. Poisson distribution

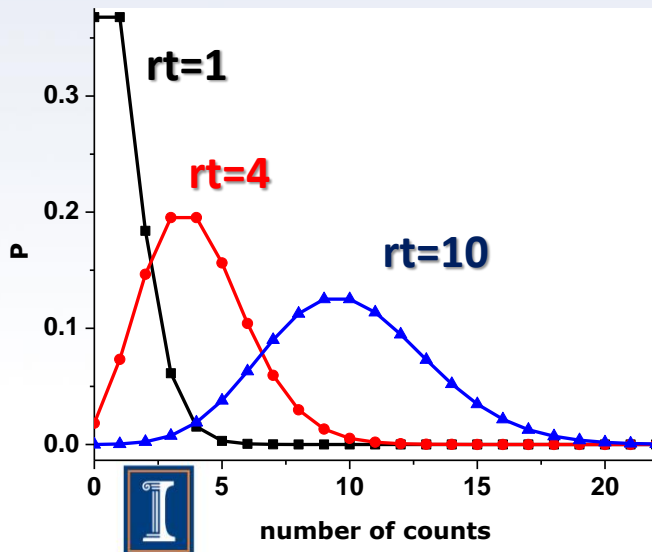


Siméon Denis Poisson
(1781-1840)

$$P_n(t) = \frac{(rt)^n}{n!} e^{-rt} \quad n = 0, 1, 2, \dots$$

r : decay rate [counts/s] t : time interval [s]

→ $P_n(rt)$: Probability to have n decays in time interval t



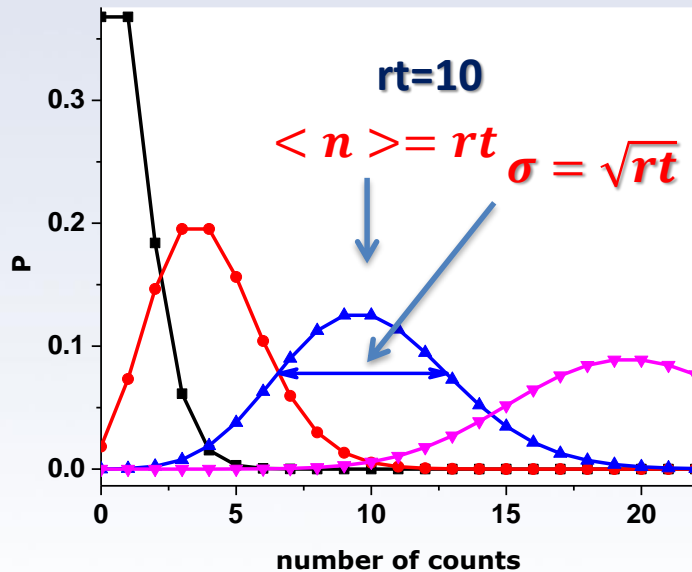
A statistical process is described through a Poisson Distribution if:

- **random process** → for a given nucleus probability for a decay to occur is the same in each time interval.
- **universal probability** → the probability to decay in a given time interval is same for all nuclei.
- **no correlation between two instances** (the decay of one nucleus does not change the probability for a second nucleus to decay).

Poisson distribution

$$P_n(t) = \frac{(rt)^n}{n!} e^{-rt} \quad n = 0, 1, 2, \dots$$

r: decay rate [counts/s] **t**: time interval [s]
 → $P_n(rt)$: Probability to have n decays in time interval **t**



Properties of the Poisson distribution:

$$\sum_{n=0}^{\infty} P_n(rt) = 1, \text{ probabilities sum to 1}$$

$$\langle n \rangle = \sum_{n=0}^{\infty} n \cdot P_n(rt) = rt, \text{ the mean}$$

$$\sigma = \sqrt{\sum_{n=0}^{\infty} (n - \langle n \rangle)^2 P_n(rt)} = \sqrt{rt}, \text{ standard deviation}$$



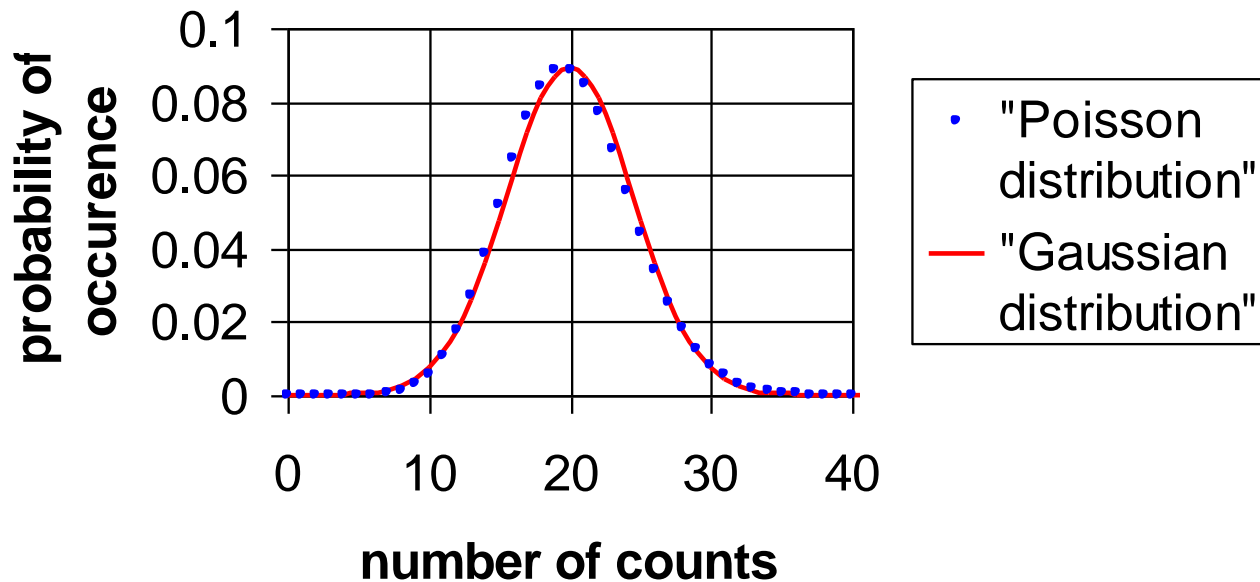
Poisson distribution at large rt

$$P_n(t) = \frac{(rt)^n}{n!} e^{-rt} \quad n = 0, 1, 2, \dots$$



**Carl Friedrich Gauss
(1777–1855)**

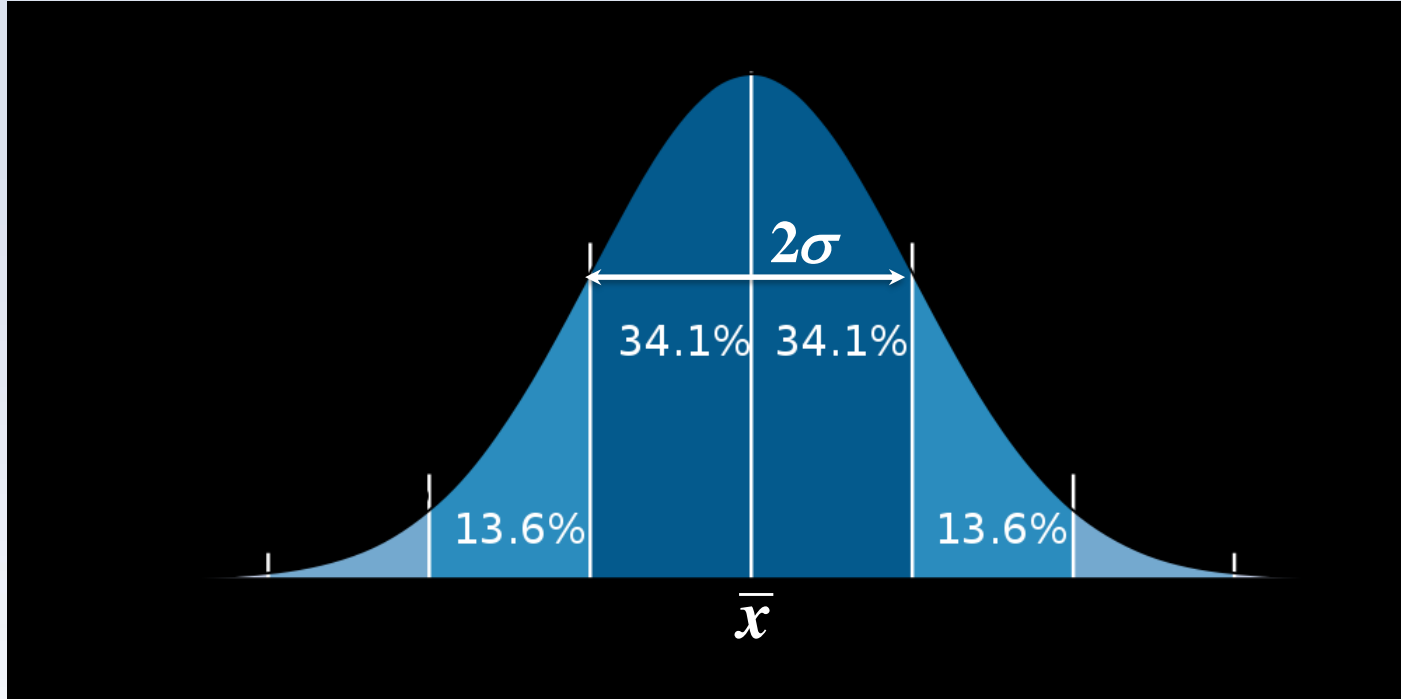
Poisson and Gaussian distributions



$$P_n(x) = \frac{1}{\sigma\sqrt{2\pi}} e^{-\frac{(x-\bar{x})^2}{2\sigma^2}}$$

**Gaussian distribution:
continuous**

Normal (Gaussian) distribution

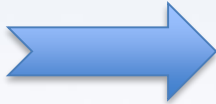
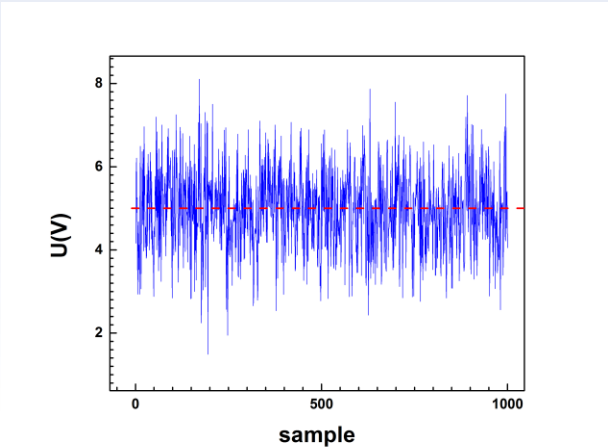
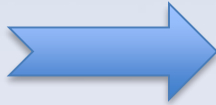


$$P_n(x) = \frac{1}{\sigma\sqrt{2\pi}} e^{-\frac{(x-\bar{x})^2}{2\sigma^2}}$$



Measurement in presence of noise

Source of noisy signal



- 4.89855
- 5.25111
- 2.93382
- 4.31753
- 4.67903
- 3.52626
- 4.12001
- 2.93411

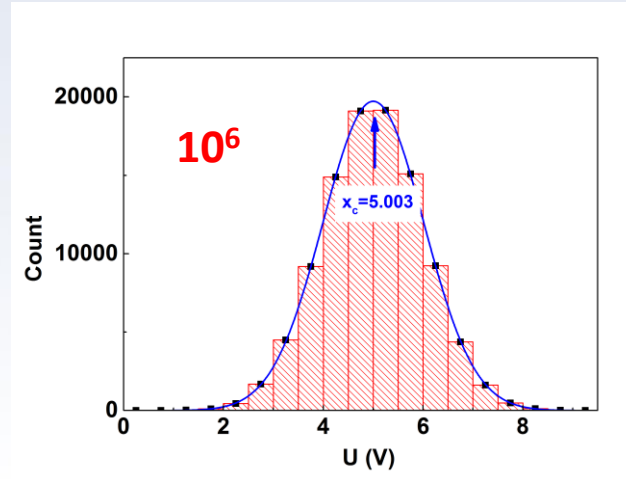
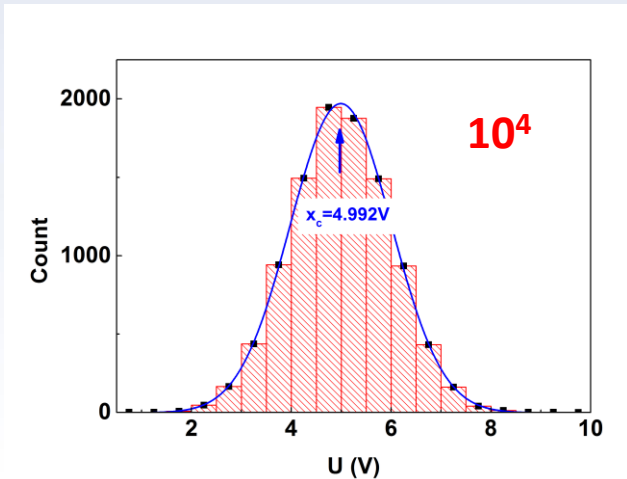
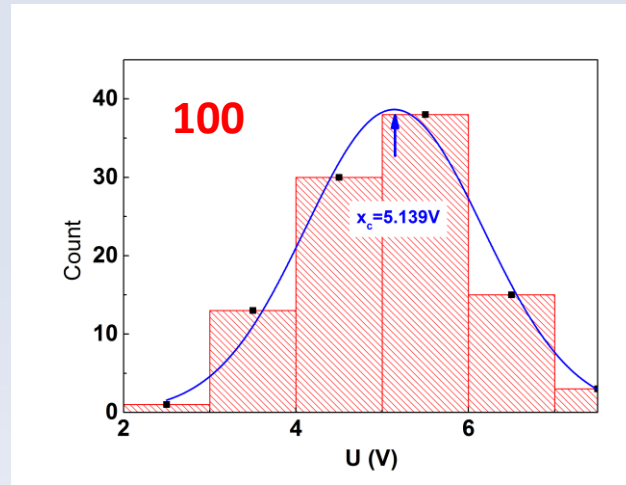
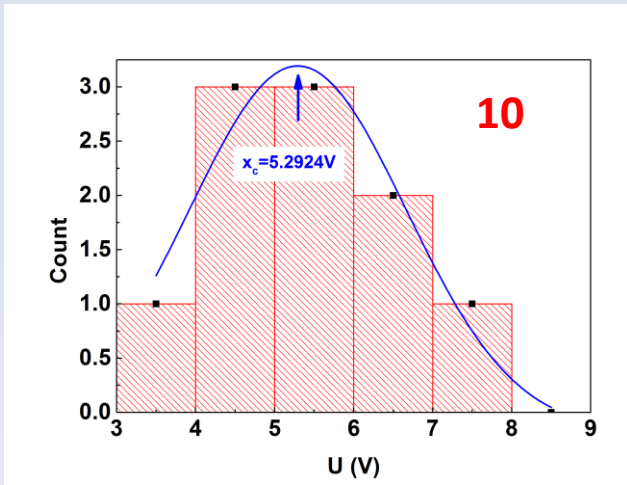
Expected value 5V



Actual measured values



Measurement in presence of noise



Error in the mean is given as $\frac{\sigma_0}{\sqrt{N}}$ (This is called standard quantum limit or the shot noise limit)



Heisenberg limit measurements

According to Heisenberg uncertainty,

the ultimate precision of the energy measurement is $\Delta E \sim \frac{\hbar}{t}$

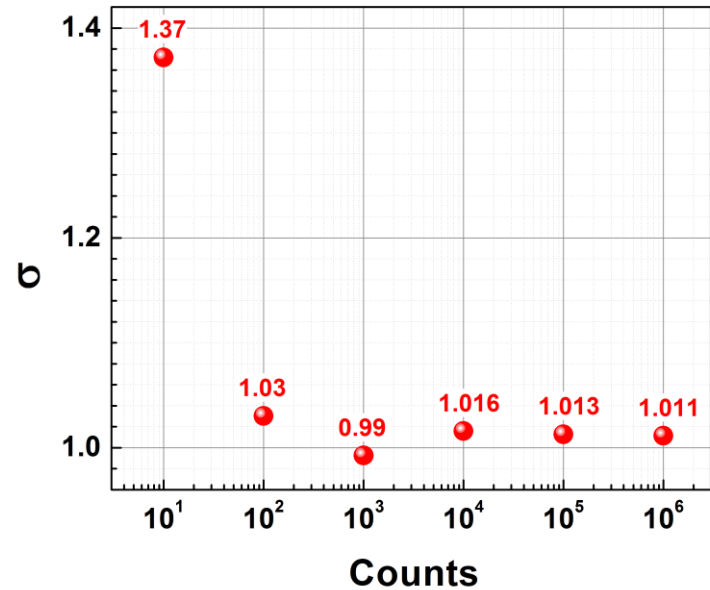
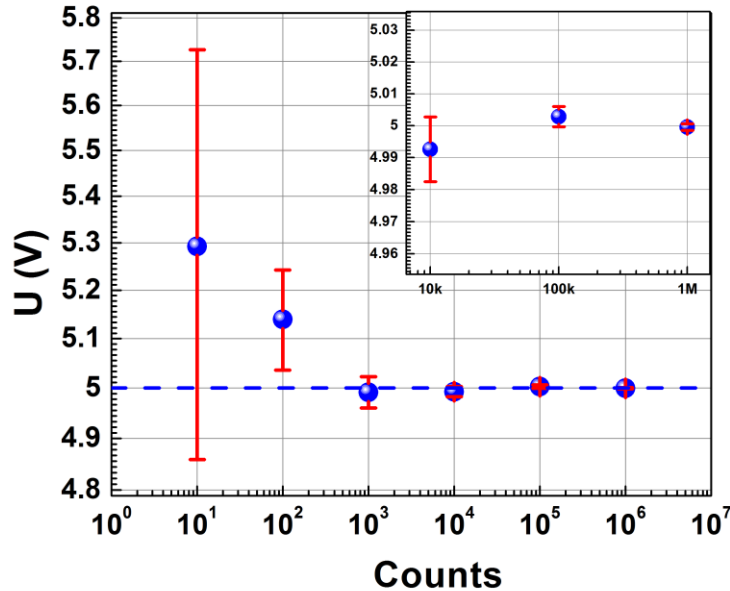
If N is the number of measurements performed then $t = N \cdot t_1$, where t_1 is the time needed to perform one measurement.

Thus the precision can be as good as $\Delta E \sim \frac{\hbar}{t_1} \frac{1}{N}$

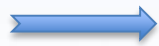
To achieve this high precision one has to use a quantum system, such as a **qubit**.



Measurement in presence of noise



Result



$$U = x_c \pm \frac{\sigma}{\sqrt{N}}$$

σ - standard deviation
 N - number of samples

For $N=10^6$ $U=4.999 \pm 0.001$

0.02% accuracy

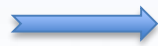


Measurement in presence of noise

The standard error equals the standard deviation divided by the square root of the sample size (=number of measurements).

In other words, the standard error of the mean is a measure of the dispersion of sample means around the population mean.

Result



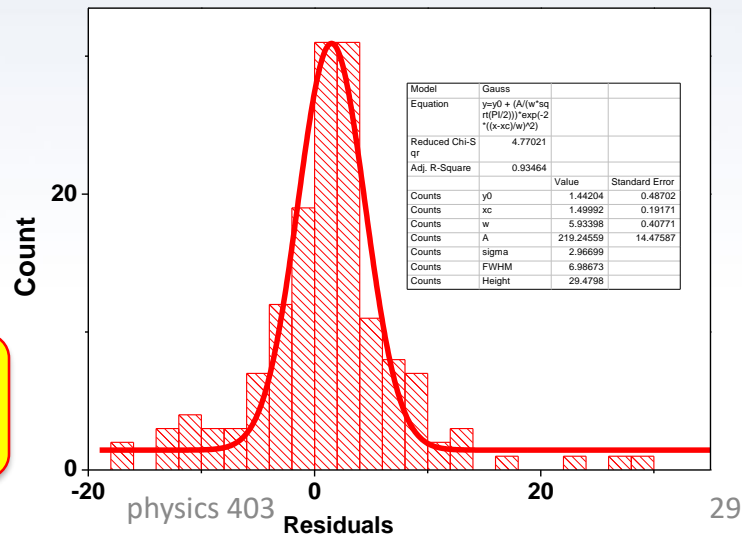
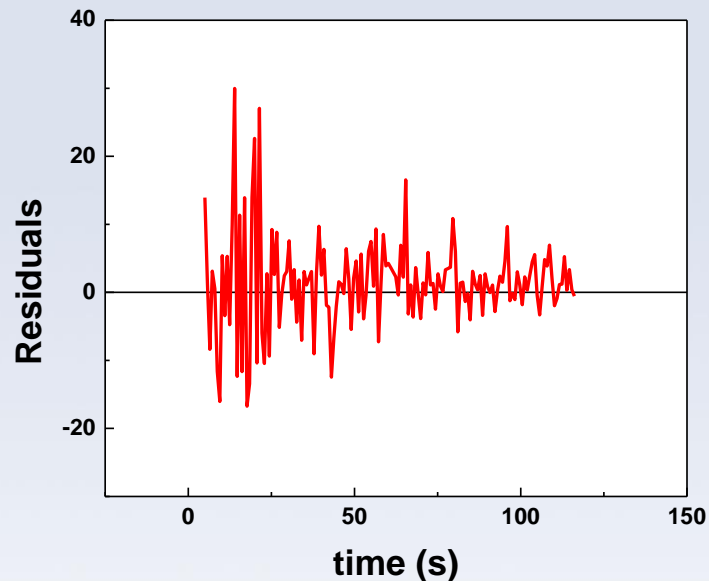
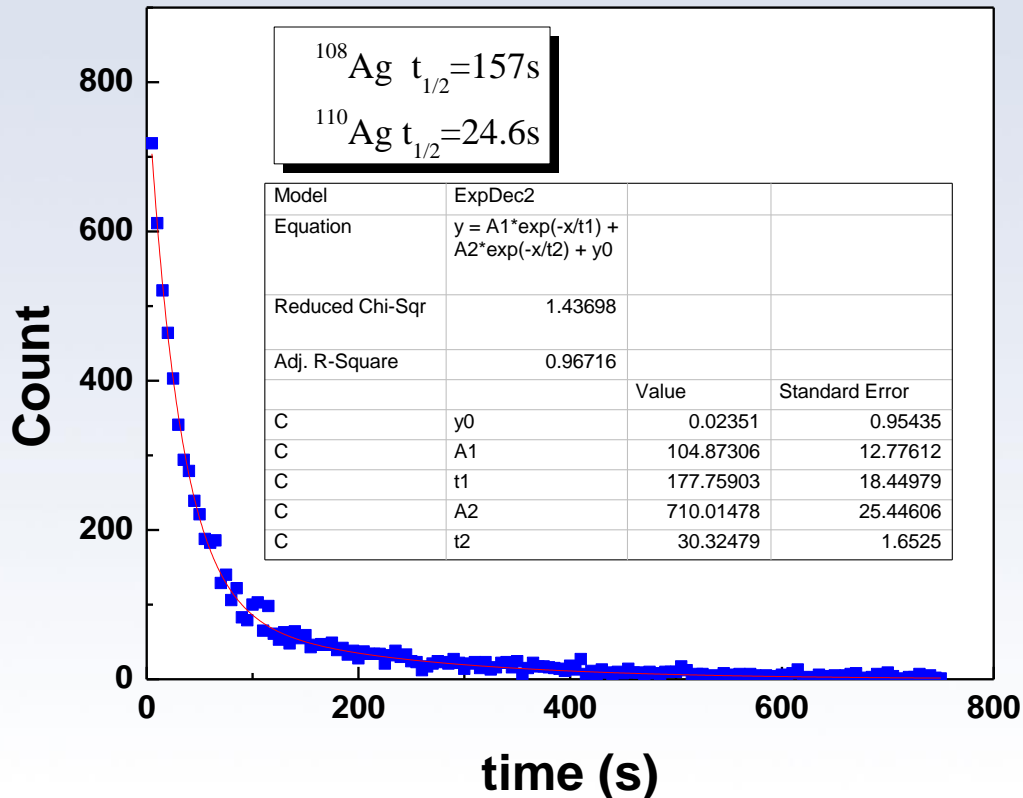
$$U = x_c \pm \frac{\sigma}{\sqrt{N}}$$

σ - standard deviation
 N - number of samples



Fitting errors

Ag β decay

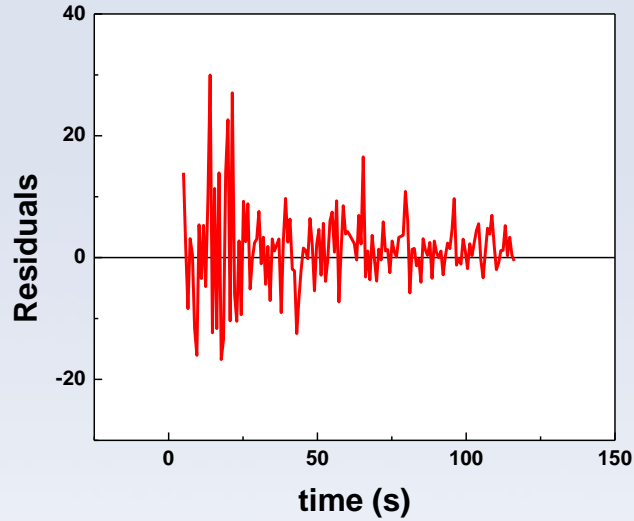


$$y = A1 \cdot \exp\left(\frac{-t}{t_1}\right) + A2 \cdot \exp\left(\frac{-t}{t_2}\right) + y_0$$

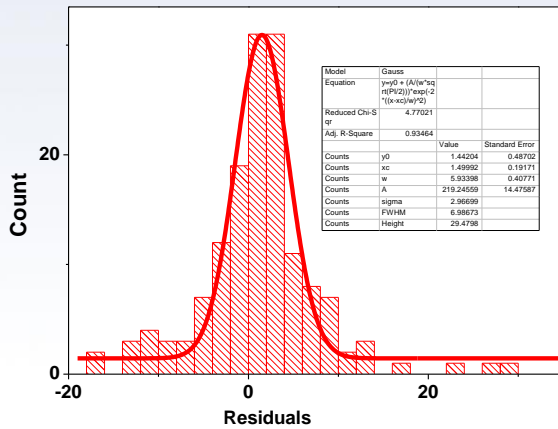
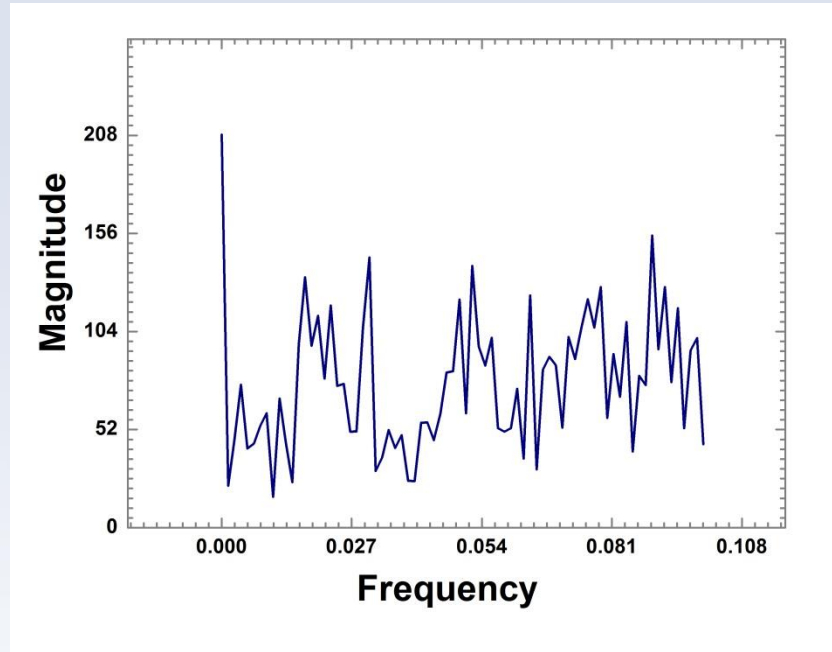


Fitting. Analysis of the residuals

Ag β decay



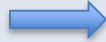
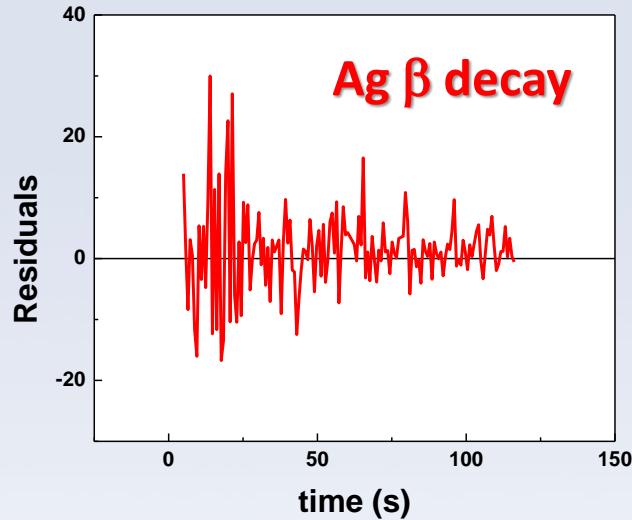
Test 1. Fourier analysis



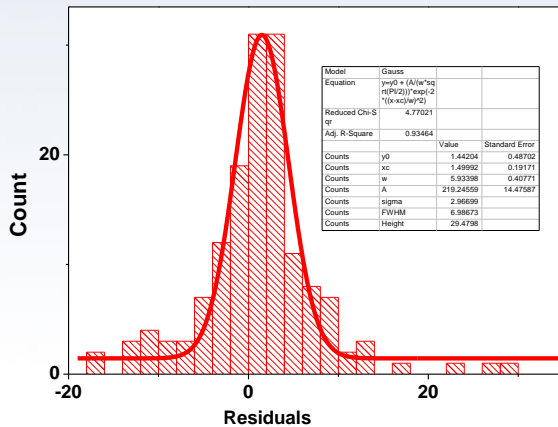
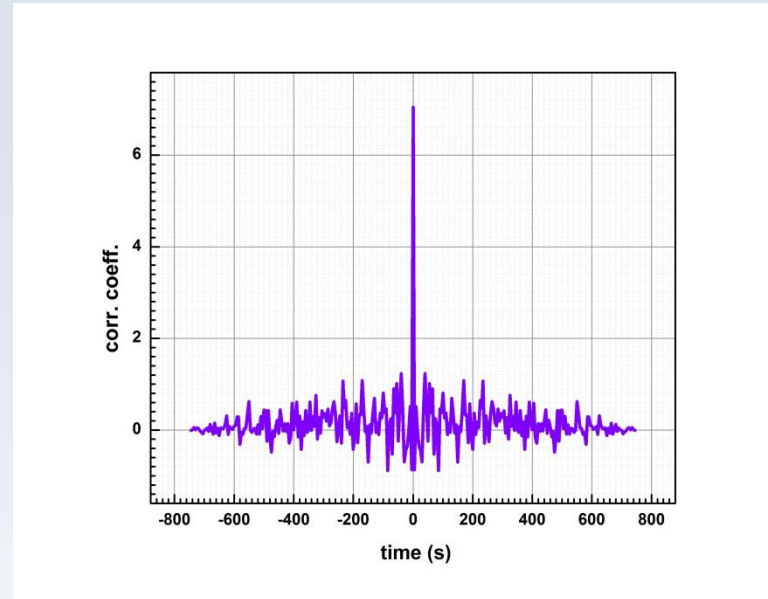
No pronounced frequencies found



Fitting. Analysis of the residuals



Test 1. Autocorrelation function



Correlation function

$$y(m) = \sum_{n=0} f(n)g(n-m)$$

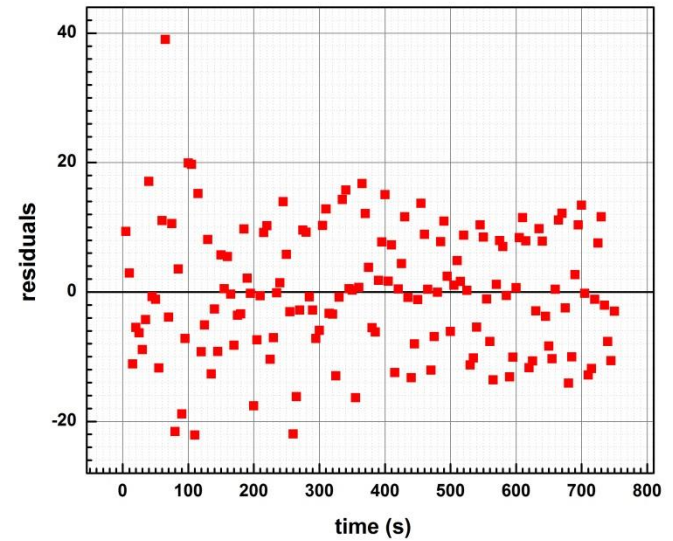
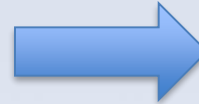
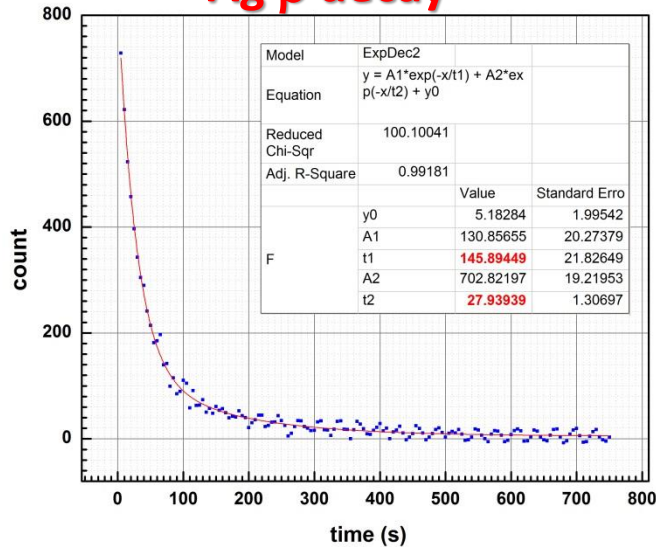
autocorrelation function

$$y(m) = \sum_{n=0}^{M-1} f(n)f(n-m)$$



Fitting. Analysis of the residuals. Non "ideal" case

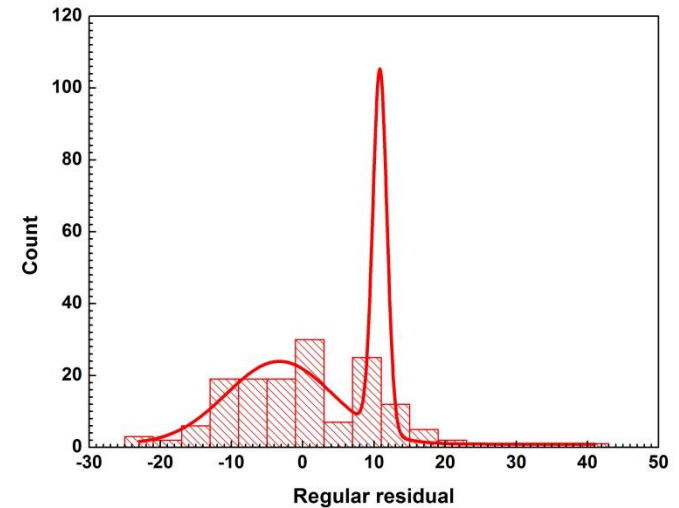
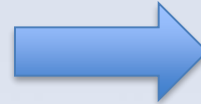
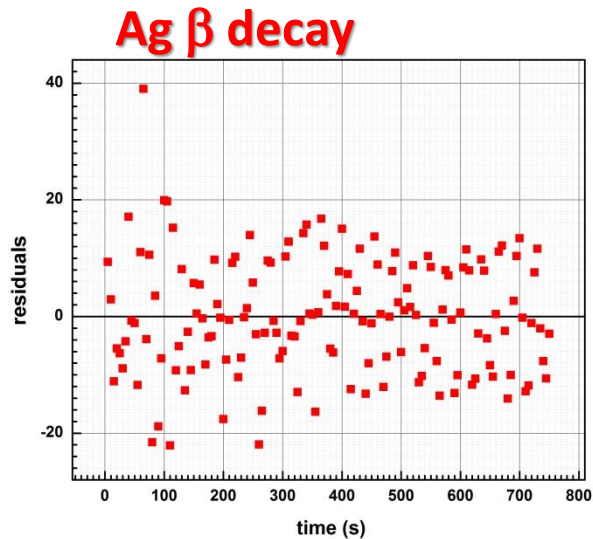
Ag β decay



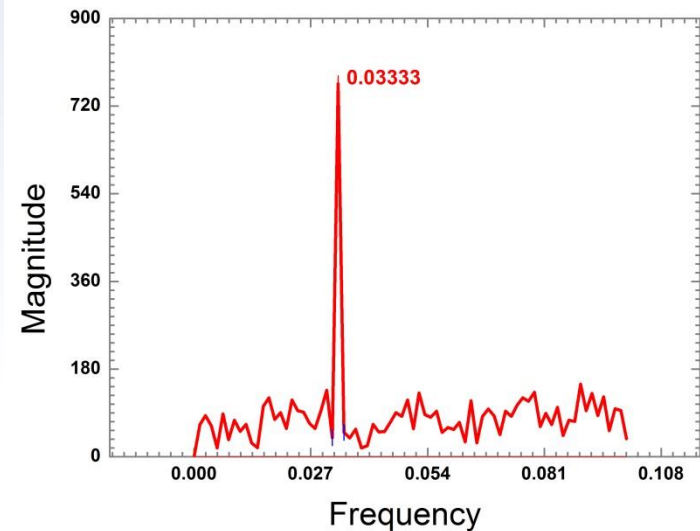
	Clear experiment	Data + "noise"
$t_1(s)$	177.76	145.89
$t_2(s)$	30.32	27.94



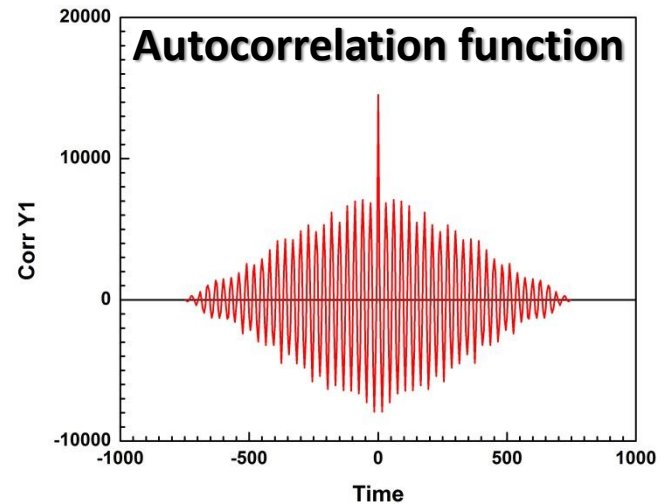
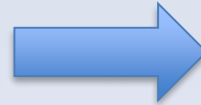
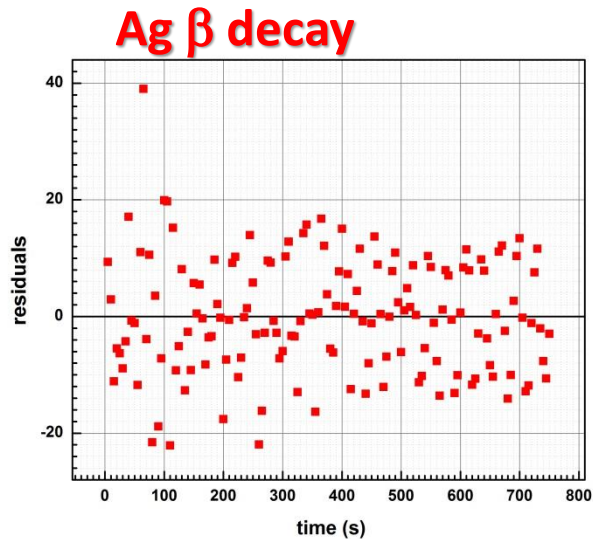
Fitting. Analysis of the residuals. Non "ideal" case



Histogram does not follow the normal distribution and there is frequency of 0.333 is present in spectrum



Fitting. Analysis of the residuals. Non "ideal" case

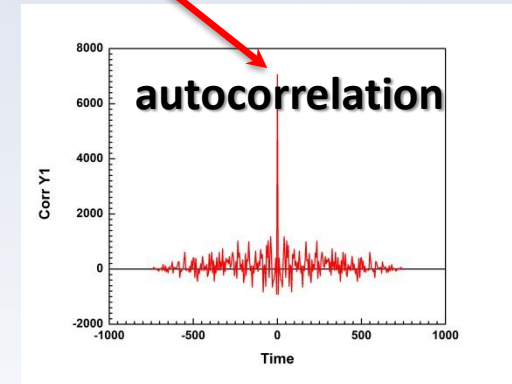
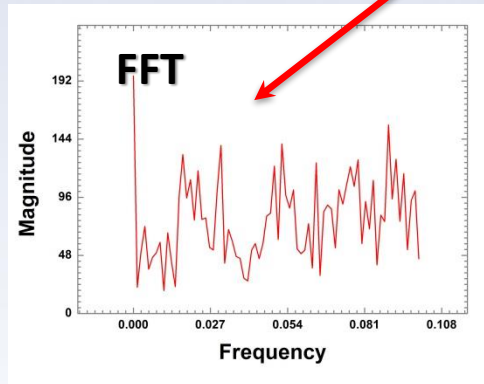
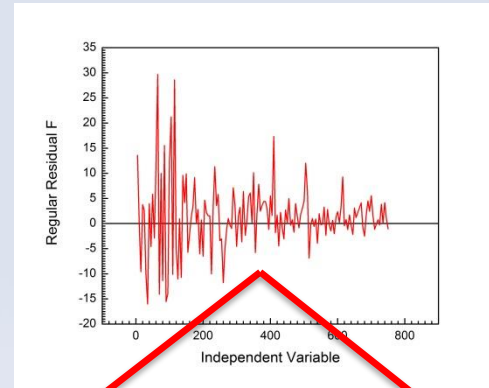
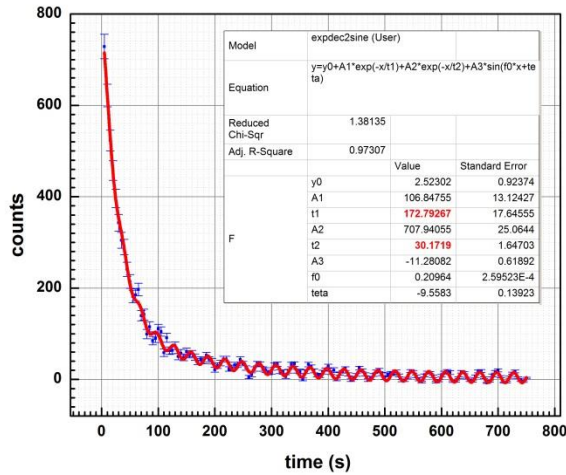


Conclusion: fitting function should be modified by adding an additional term:

$$y(t) = y_0 + A_1 \exp\left(\frac{-t}{t_1}\right) + A_2 \exp\left(\frac{-t}{t_2}\right) + A_3 \sin(\omega t + \theta)$$



Fitting. Analysis of the residuals. Non "ideal" case



	Clear experiment	Data + noise	Modified fitting
$t_1(s)$	177.76	145.89	172.79
$t_2(s)$	30.32	27.94	30.17



Error propagation. Example.

Derive resonance frequency f
from measured inductance
 $L \pm \Delta L$ and capacitance $C \pm \Delta C$

$$f(L, C) = \frac{1}{2\pi} \sqrt{\frac{1}{LC}}$$

$$L_1 = 10 \pm 1 \text{mH}, \quad C_1 = 10 \pm 2 \mu\text{F}$$

$$\Delta f(L, C, \Delta L, \Delta C) = \sqrt{\left[\frac{\partial f}{\partial L}\right]^2 \cdot \Delta L^2 + \left[\frac{\partial f}{\partial C}\right]^2 \Delta C^2}$$

$$\frac{\partial f}{\partial L} = \frac{-1}{4\pi} C^{-\frac{1}{2}} L^{-\frac{3}{2}};$$

$$\frac{\partial f}{\partial C} = \frac{-1}{4\pi} L^{-\frac{1}{2}} C^{-\frac{3}{2}}$$

Results:

$$f(L_1, C_1) = 503.29212104487 \text{Hz}$$

$$\Delta f = 56.26977 \text{Hz}$$

$$f(L_1, C_1) = 503.3 \pm 56.3 \text{Hz}$$



Presentation of the results. Reports/publications. Condensed Matter Physics.

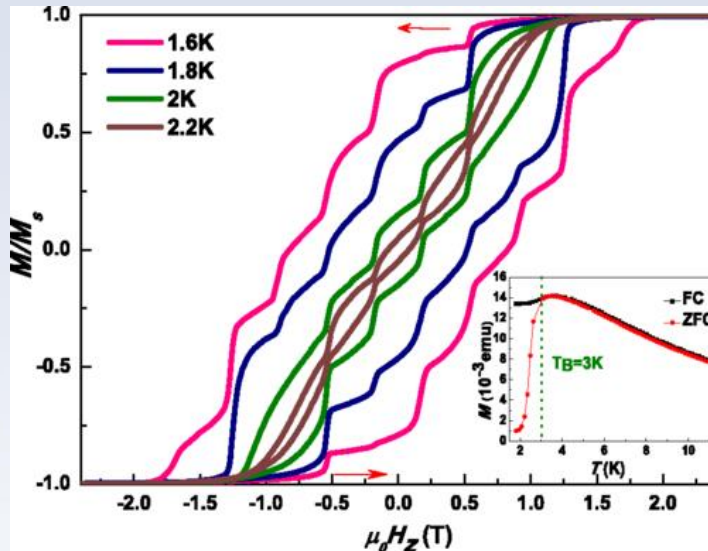


Figure 3. Magnetization (M/M_s) of Mn3 single crystal versus applied magnetic field with the sweeping rate of 0.003 T/s at different temperatures. The inset shows ZFC and FC curves.

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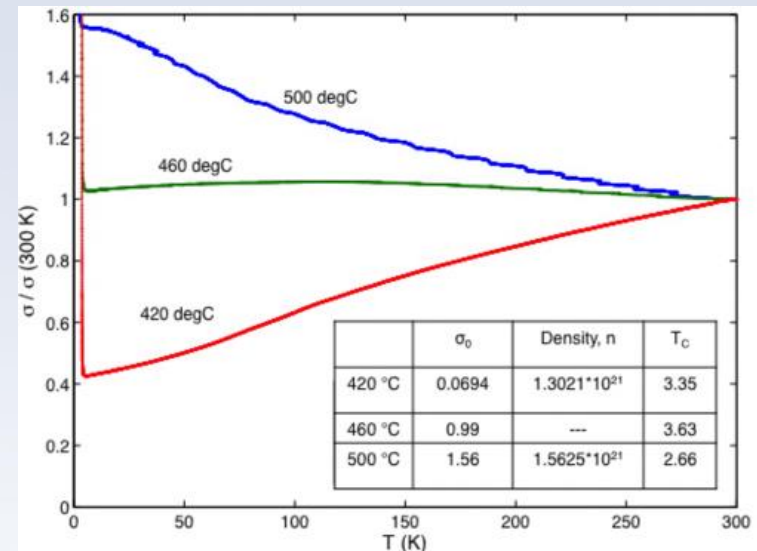


Figure 2. Normalized conductivity vs temperature for three 250-nm-thick $K_{0.33}WO_{3-y}$ films on YSZ substrates. The films are annealed in vacuum at different temperatures, with properties shown in the inset table. The units of T_{anneal} are degrees Celcius, σ_0 is given in $1/m\Omega cm$, n in $/cm^3$, and T_c in degrees Kelvin.

Phys. Rev. B 89, 184501

Presentation of the results. Reports/publications. Particle Physics.

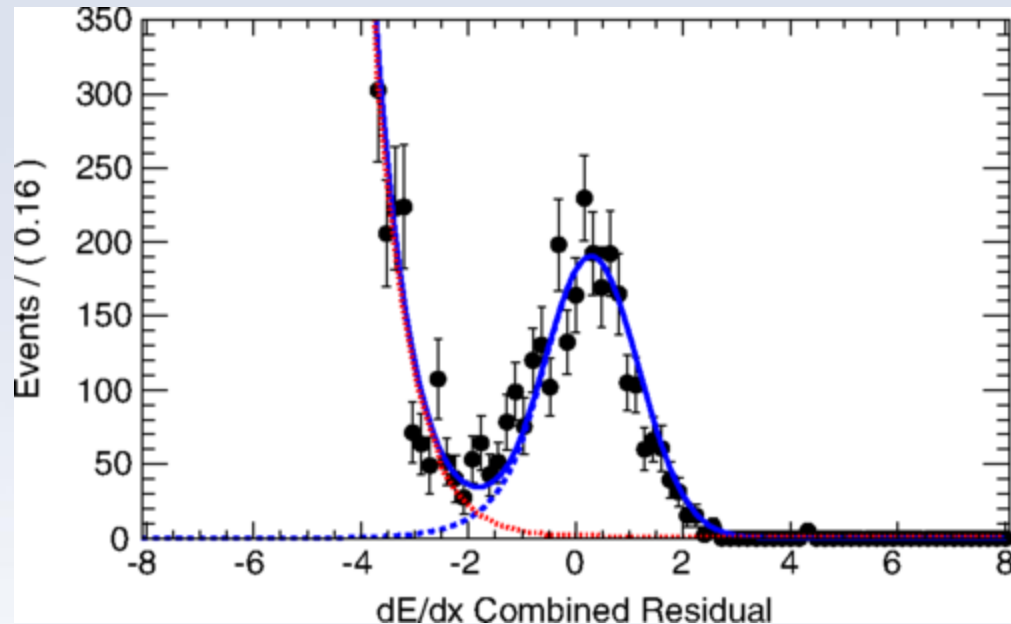


Figure 1. Normalized residuals of the combined dE/dx for antideuteron candidates in the Onpeak $\Upsilon(2S)$ data sample, with fit PDFs superimposed. Entries have been weighted, as detailed in the text. The solid (blue) line is the total fit, the dashed (blue) line is the d^- signal peak, and the dotted (red) line is the background.

Phys. Rev. D 89, 111102(R)



Presentation of the results. Student Reports.

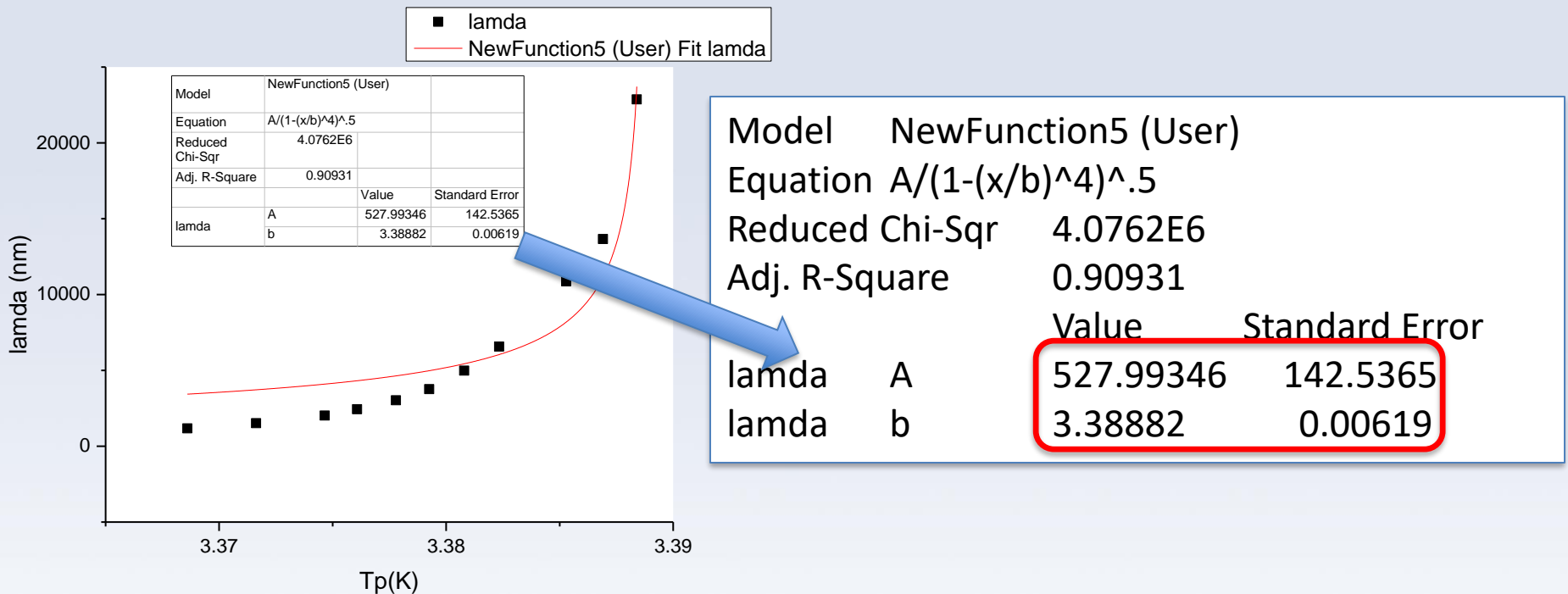


Figure 10(ii): lambda versus T for indium film with thickness 300 nm. Input voltage is 0.2v. Critical temperature(b) and penetration depth(A) at temperature 0 K is determined

Spring 2014.



Presentation of the results. Student Reports.

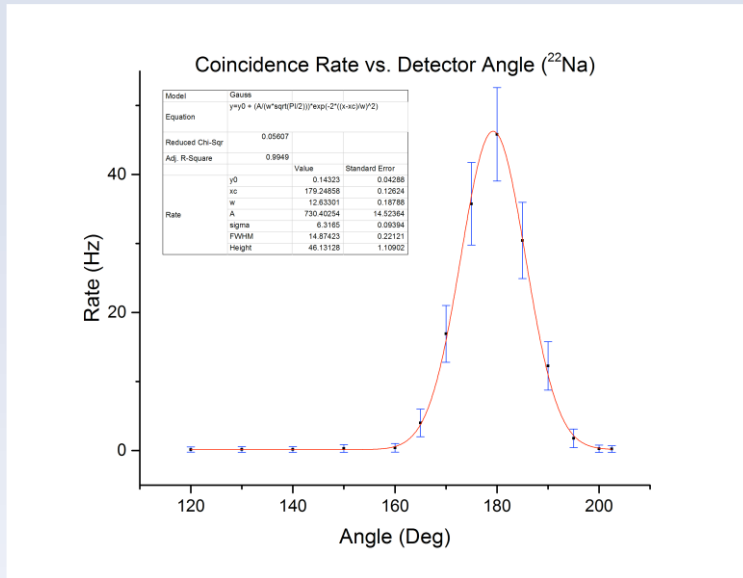


Figure 8: Coincidence Rate vs. Detector Angle for ^{22}Na correlation measurement.

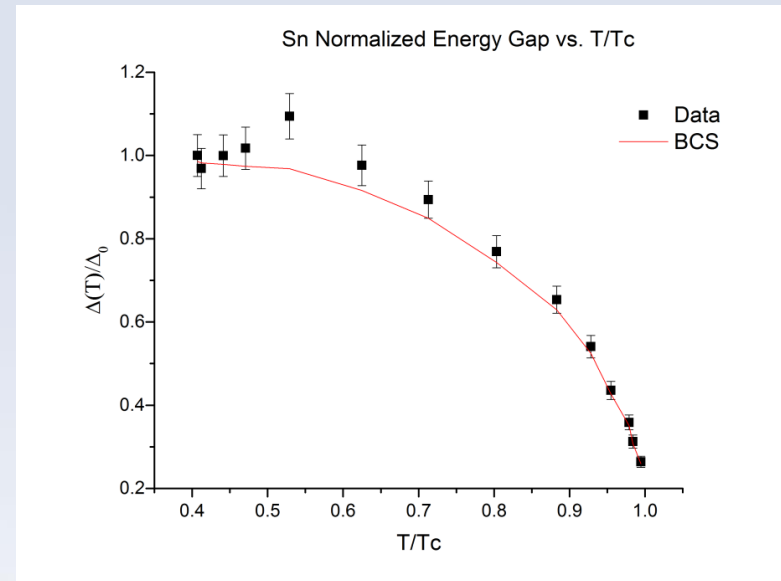


Figure 11: Temperature dependence of energy gap in Sn. Red line is BCS theory

Summer 2019.

