

Small-Size Dilution Cryostat for Rapid Cooling

Jaakko Koivuniemi

Low Temperature Laboratory, Helsinki University of Technology,
P.O. Box 2200, FIN-02015 HUT, Finland

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Abstract

A small test dilution cryostat has been constructed. Reliable operation with base temperature of 50 mK was obtained. Design considerations and performance are discussed. Cool down from room temperature to 100 mK takes 3 – 5 hours.

1 Introduction

Ultralow temperature experiments are usually performed in large cryostats for several reasons: efficient heat exchangers of dilution cryostat with large cooling power require some volume and nuclear cooling with copper demands large magnetic fields of ~ 6 T and a big nuclear stage. Such large cryostats have a long cool down time and require careful testing of vacuum leaks and gas flow within the cryostat. Recently there has been increasing interest in small dilution cryostats capable of reaching a base temperature of 10 – 50 mK [1, 2, 3]. These have fast turn around time and are well suited for testing different devices or for checking out equipment for experiments which will later be conducted in a big nuclear demagnetisation cryostat. It is likely that most practical problems will be uncovered by such testing in the small dilution machine. For instance, measured signals and thermal noise are on a more realistic scale than in more simple tests at 1.5 – 4 K in liquid ^4He dewar.

2 Operating principle of a dilution cryostat

A mixture of ^3He and ^4He liquids phase separates into two parts below 0.87 K [4, 5, 6], Fig. 1. The upper volume consists of the ^3He rich phase and the lower volume of a dilute solution of ^3He in superfluid ^4He . The concentration of ^3He in the upper volume approaches unity when $T \rightarrow 0$ K. In the diluted solution the concentration

of ^3He stays finite and approaches 6% at zero temperature. When ^3He atoms move from the ^3He rich phase to the diluted phase cooling is observed. The cooling power is due to the difference in the enthalpy of the diluted and concentrated ^3He phases

$$\dot{Q}_M = \dot{n}_3(96T_M^2 - 12T_N^2), \quad (1)$$

where \dot{n}_3 is the amount of ^3He atoms crossing the phase boundary in mol/s, T_M the mixing chamber temperature, and T_N the temperature of the incoming ^3He . For zero heat load the minimum temperature is determined by the temperature of the incoming ^3He stream $T_M \approx T_N/2.8$.

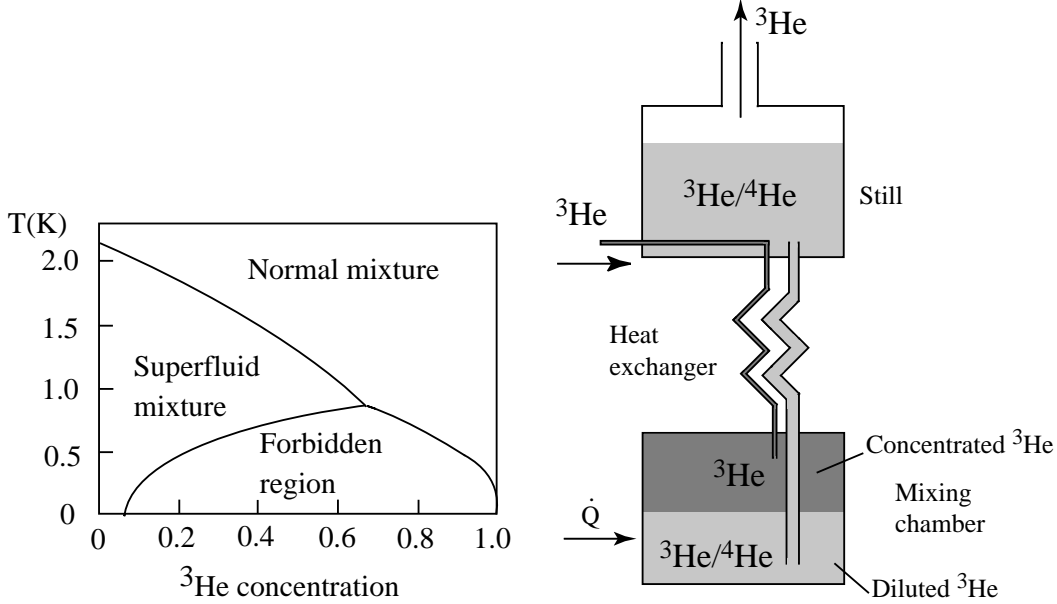


Figure 1: *Left:* Phase diagram of $^3\text{He}/^4\text{He}$ mixtures. The molar concentration of ^3He in the mixture $n_3/(n_3 + n_4)$ is plotted on the horizontal axis. Below 0.87 K the mixture phase separates into two different volumes: a ^3He rich phase floats on top of a diluted $^3\text{He}/^4\text{He}$ phase. At zero temperature the concentration of ^3He in the upper volume approaches unity while for the lower volume it becomes 6.4 %. *Right:* Principle of a dilution cryostat. The cooling is produced in the mixing chamber where ^3He atoms move from the ^3He rich phase to the diluted $^3\text{He}/^4\text{He}$ phase. The ^3He flow is maintained by evaporating the ^3He from the diluted solution in the still. The pumped gas is returned back into the cryostat and condensed to liquid before it enters the heat exchanger in the still.

In a practical dilution cryostat the phase boundary between the diluted ^3He and concentrated ^3He is located in the mixing chamber. The incoming ^3He is cooled in a heat exchanger. A second vessel above the mixing chamber, the still, is used for evaporating ^3He . The pumped gas is fed back to the cryostat and condensed to liquid at the ^4He evaporation cooler, the ^4He pot.

3 Design considerations

The cryostat should be easy to use with fast cool down time. It should have a modular structure to allow easy modification for different experiments. There should be space

for many shielded wires, GaAs MESFET amplifier, DC SQUID with its shields and for an experiment below the mixing chamber. A 1 K plate with a ^4He pot is needed. It serves as the first thermal anchor for the wires inside the cryostat. It also provides a fixed operating temperature for the DC SQUID when the liquid ^4He level in the dewar is already below the flange of the vacuum jacket. The vacuum can and pot flanges should have the provisions to fix different accessories on them.

3.1 ^4He pot - design and performance

The pot design is fairly simple: it is a copper vessel soldered on the pot flange, Fig. 3. A 1.6 mm diameter CuNi tube connects the pot volume to a larger pumping tube. It works as a superfluid He-II film stopper at the same time. In the pot the returning ^3He is condensed to liquid in a coiled copper tube. The pot is continuously filled with a syphone extending to the liquid helium bath outside. The temperature of the pot is measured with a RuO_2 chip resistor of nominal value of 1 k Ω (type RCWPM-575 from Dale Electronics).

The main optimization parameter for the pot is the flow impedance from the liquid helium bath. The pot should not run empty too easily and does not need to maintain very low temperature. Thus the impedance can be of fairly low value, 10 - 20 cm of 0.1 mm inner diameter CuNi tube is usually sufficient. The tall shape of the pot gives some reserve to sudden heat loads (like in starting the condensation of mixture or circulation in the dilution machine). In stable operation of the pot the liquid level inside it is high, close to the 1.6 mm CuNi pumping tube. This can be seen in the small oscillations in the base temperature of the pot: When the liquid level reaches small pumping tube, the evaporating and cooling power are suddenly decreased and the pot warms up. It cools down again when the liquid level drops so that there is a larger evaporating area. This was tested with different impedances. First a small impedance was used and the oscillations in the pot temperature started immediately when the pumping was started. The reason for this was the rapid filling of the pot. With a larger impedance the temperature was stable in the beginning and started oscillating between 1.68 - 1.70 k Ω after 30 minutes of pumping.

The pot cooling power was tested with a heater, Fig. 2. First the pot was filled with liquid ^4He . The heat load was gradually increased and a small drop in the thermometer resistance was observed. The heating was kept constant for five minutes. After the last point with 50 mW of heating the pot ran empty. This was seen as a sudden drop in the thermometer resistance to 1250 Ω . Thus with the gradually increasing heating the pot was empty in 30 minutes. When the heating was decreased the thermometer resistor followed a different curve and the pot stayed at a relatively high temperature. Below 10 mW heating it started to fill with liquid again.

3.2 Still

The still is pumped via a 8 mm diameter tube. There is a sharp knife-edge orifice in the pumping channel to restrict the ^4He superfluid film from creeping up to a high temperature point in the pumping tube and evaporating there. At the knife edge there is a large gradient in the pressure. Part of the ^4He film will evaporate there. The rest of the film flow is reduced due to the small surface area of the knife-edge. The ^4He circulation is a problem because it does not participate in the cooling of

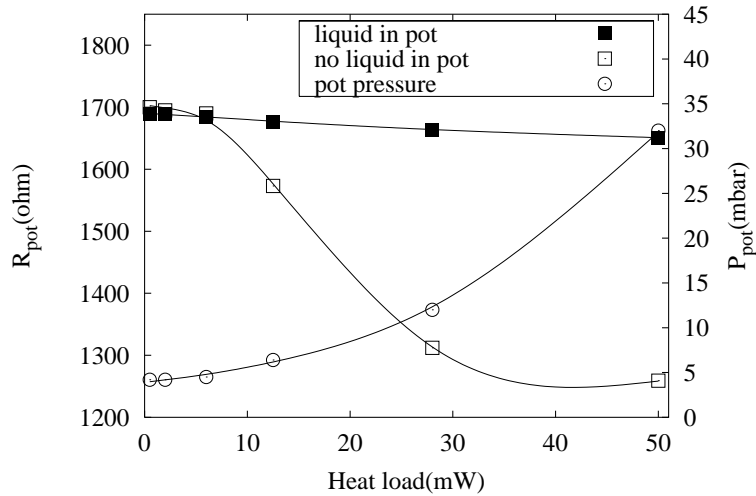


Figure 2: The pot temperature at different heat loads.

the dilution refrigerator but loads the pumping system [5]. The restriction is in practice, however, a sufficient precaution for a small dilution machine. The still has an indium sealed bottom. To this bottom part four copper feedthroughs have been hardsoldered. It is easy to use these feedthroughs by soldering a CuNi tube into them with ordinary soft solder. The CuNi tube contains either the circulating ^3He or wires going into the still. The wires are sealed to this tube with the standard method by using a feedthrough molded from Stycast 1266 epoxy [4, 6]. The temperature of the still is measured with a RuO_2 resistor of nominal value $1\text{ k}\Omega$. The heater is made from twisted CuNi wire wound inside on the bottom plate of the still. The incoming ^3He is thermalised to the still temperature with a copper block fixed with a screw to the bottom. A second flow impedance (5 - 10 cm of 0.1 mm inner diameter CuNi tube) before the continuous heat exchanger ensures that no condensation of re-evaporated ^3He happens in the still, or worse, in the heat exchanger.

3.3 Continuous heat exchanher

The continuous heat exchanger is made from 1.6 mm diameter CuNi tube with a PTFE tube of 1 mm diameter inside it. The connection between the second flow impedance and the Teflon tube is sealed with vacuum grease. The thermal contraction of the Teflon helps in making this joint leak tight. On the otherhand, even if there is a small leak, it should not deteriorate the performance too much. The plastic has a small density compared to the usual metallic heat exchangers. Thus the phonon mismatch, giving rise to Kapitza boundary resistance [7] for heat flow between solid and liquid, is smaller. The CuNi tube is wrapped on a Teflon frame mechanically connecting the still and the mixing chamber. This frame gives a fairly large, of order of $1\text{ }\mu\text{W}$, heat leak to the mixing chamber.

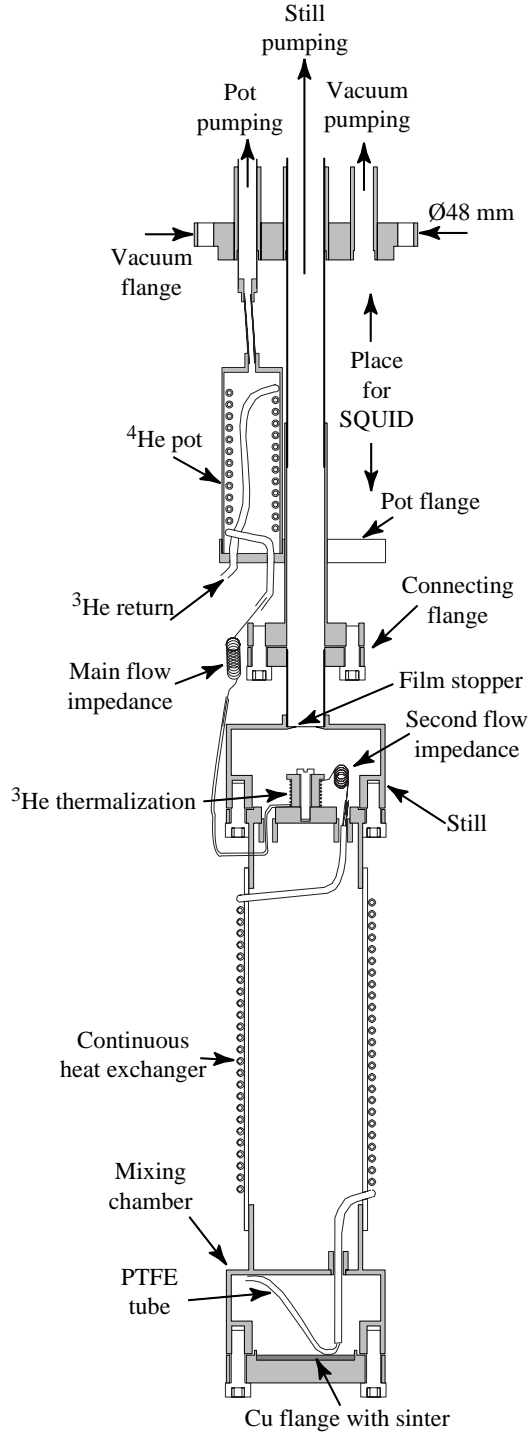


Figure 3: Construction of the dilution cryostat. The vacuum flange on the top has three pumping tubes for vacuum, pot and still. The ^4He evaporation cooler or pot gives the 1.5 K temperature that is needed for condensing incoming ^3He (and the ^4He in the initial filling) of the dilution unit. The main flow impedance is a 20 cm long, 0.1 mm inner diameter CuNi tube. The still pumping tube is connected with a indium sealed flange to the still. Thus the unit can be easily removed. The still can also be opened easily. A teflon frame connects mechanically still to mixing chamber. A simple continuous heat exchanger with a CuNi outer and PTFE inner tube is used. The experiment is fixed to the mixing chamber flange.

3.4 Mixing chamber

The mixing chamber has the same volume as the still. It is made from stainless steel AISI 316L with a bottom flange made from copper and sealed with an indium O-ring joint. The Cu-flange has silver powder sintered on it. There are threads on the Cu-flange for fixing thermometers, heaters and the experiment. The Teflon tube that carries the incoming ^3He goes to the concentrated part of the mixture. The diluted fraction of the solution is pumped through the annular volume between the Teflon tube and the CuNi tube.

4 Cooldown from room temperature

The vacuum jacket of the cryostat is sealed with an indium O-ring joint. It is pumped through the vacuum pumping tube. Usually the valves to the dilution machine are kept closed when the dilution machine is warm. Thus there is no need to pump the dilution machine. The same is true for the ^4He pot: After the previous cooldown it is filled with an overpressure of gaseous ^4He . The overpressure is leaking slowly from the pot through the syphone. No air or moisture will be flowing to the pot which would block the flow impedance capillary during cool down. Before cooldown the pot is pressurised with 1.5 atm of helium. Thus there will be a small overpressure in the pot during the cool down to the liquid nitrogen temperature.

The cryostat is inserted into liquid nitrogen vessel for precooling. By monitoring the vacuum pressure it is possible to see if there are any leaks. The pressure should decrease monotonically. A fairly accurate pressure gauge with digital display is useful in monitoring such small changes in the pressure. Only if a leak is observed is it necessary to use a leak detector to determine where the leak is. Most leaks are observed already at LN_2 temperature. Heat exchange gas is necessary to cool the inner parts of the cryostat. The gas can be either air, neon or ^4He . The precooling with 1 – 5 mbar of helium is fast and takes about 10 – 15 minutes. Neon is used if there is need for leak detection of the dilution unit or experimental cell at LHe temperature. Precooling with air is slow and it is seldomly used.

After the cryostat has cooled to LN_2 temperature it is moved into the ^4He storage dewar. The helium heat exchange gas works also at LHe temperature and it is not necessary to pump it finally. If the amount of heat exchange gas was right in the begining it will be effectively cryopumped by the cold walls of the cryostat. Thus it does not necessarily need to be pumped away as the cryostat can operate with smaller than $5 \cdot 10^{-3}$ mbar pressure of helium in the vacuum jacket. The pot fills automatically from the liquid ^4He bath when the temperature goes down. Its operation is started by pumping through the pot pumping tube. The pot operation temperature of about 1.5 K is reached almost immediately. If there are any problems with pot operation they become now visible: Too low a temperature means that the syphone or impedance has been partially blocked by air and a superleak in the pot will be seen clearly as an increase in the pressure of the vacuum jacket.

When the proper ^4He pot operation has been confirmed, condensation of the $^3\text{He}/^4\text{He}$ -mixture into the dilution unit can be started. This is done simply by fully opening tank volume into the still pumping tube. The pot thermometer resistor will collapse to a lower value in about 15 – 30 minutes indicating that the pot has run empty of liquid ^4He . When it recovers in 1 – 2 hours time most of the condensation

has taken place and further condensation proceeds very slowly. At this point it is better to start the circulation and to condense the rest of the mixture to the dilution unit while circulating gas through the refrigerator. The still valve is opened slowly and the still is pumped for 30 minutes before the ^3He return line to the cryostat is opened. This way one makes sure that all the air goes into the LN_2 trap and does not block the cold ^3He return line. The needle valve from the storage tank is opened to the still pumping tube so that the incoming flow of ^3He does not dry the pot with too much heat load. Usually the flow is limited to below $100\text{ }\mu\text{mol/s}$. When all the gas has been added to the circulation and stable operation has been reached, then the still heater is switched on. The mixing chamber starts to cool down rapidly reaching 100 mK in about one hour.

5 Performance of the dilution cryostat

The minimum temperature in the mixing chamber, about 50 mK , was reached with about 0.7 mW heating power to the still. With larger heating the temperature started oscillating with a period of $15 - 30$ minutes. These oscillations are probably due to emptying of the still; with too much heating power the mixture in the still evaporates. The area of the heat exchanger tube connecting the still and mixing chamber is much less than the cross section area of the still. Thus, less ^3He evaporates and circulation decreases. When the extra ^3He comes back to the mixing chamber, the still is again filled with mixture and the cycle repeats itself.

The cooling power can be measured by supplying constant heat to the mixing chamber and waiting for the temperature to stabilize, Fig. 4. A heater resistor made from $100\text{ }\mu\text{m}$ CuNi wire is used for heating. The circulation was $\dot{n}_3 = 35\text{ }\mu\text{mol/s}$ and cooling power at 100 mK about $10\text{ }\mu\text{W}$. The minimum temperature is a little uncertain since the Ge-resistor (Lake Shore GR-200A-50) saturated at $1.3\text{ M}\Omega$. The thermometer is easily overheated at such high values and may indicate a higher temperature than the mixing chamber. It was calibrated against a same type factory calibrated Ge-resistor. The estimated heat leak through the teflon frame from the still to the mixing chamber is about $1.6\text{ }\mu\text{W}$. From the cooling power measurement, a $3.8\text{ }\mu\text{W}$ heat leak to the mixing chamber was obtained.

6 Conclusions

A small dilution refrigerator has been constructed which fits inside a liquid He container with a 50 mm neck and achieves reliably a base temperature of 50 mK and a cooling power of $10\text{ }\mu\text{W}$ at 100 mK . The cool down from room temperature takes $3 - 5$ hours, depending on how succesful each precooling step is. The low temperature performance of the cryostat could still be improved with better heat exchangers, e.g. with the “bellows” type step heat exchanger made from Kapton foil [7]. On the other hand a construction with soldered joints has better reliability in repeated thermal cycling than one with large plastic parts. With less ^3He in the mixture the dilution unit should cool down faster [4]. The operation is sensitive to the condensing pressure and the amount of gas in the room temperature of the pumping system: too high a pressure means that the phase boundary is not in the mixing chamber since too much ^3He is in the back of the pump. The space between the still and mixing

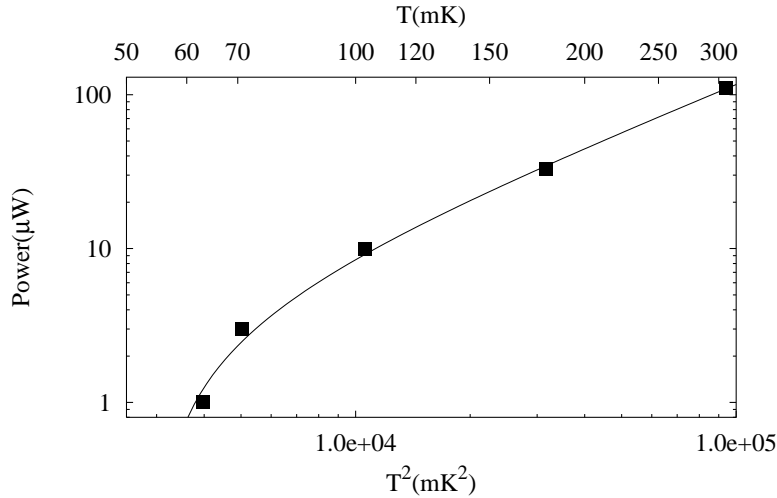


Figure 4: The mixing chamber cooling power. The heating power is plotted on the vertical axis. Below $1 \mu\text{W}$, the heating is dominated by an external heat leak. A lower base temperature could be obtained by reducing this heat leak. The line represents a fitted $\dot{Q} \propto T^2$ dependence.

chamber is now not used. With a better design of the mixing chamber this space could be used for experiments or a step heat exchanger could be placed there.

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