Physics 489 S 04 Lecture 24 Magnetism I: Diamagnetism and Paramagnetism "Local moments" in solids - due to electron-electron interactions Aschroft and Mermin, Ch. 31; Kittel Ch. 14

- 1. Magnetism: Purely quantum phenomenon. Bohr-van Leeuwen Theorem $B = H + 4\pi M$ in cgs units Magnetization $M = -\frac{1}{V}\frac{dF}{dH}$, F = free energy Magnetic susceptibility $\chi = dM/dH = -\frac{1}{V}\frac{d^2F}{dH^2}$
- 2. Interaction of electrons with magnetic field Orbital motion: $\mathbf{p} \to \mathbf{p} - (-e/c)\mathbf{A}$ Spin: $g_0\mu_B(\mathbf{H}\cdot\mathbf{s})$
- 3. Pauli spin paramagnetism (plus Landau current diamagnetism) in metals $\chi_{Pauli} = \mu_B^2 g(\epsilon_F) \approx 3(N/V)(\mu_B^2/2\epsilon_F)$ In independent electron approximation $C_V/T = (\pi^2/3)k_B^2 g(\epsilon_F)$ Landau current diamagnetism: $\chi_{Landau} = -(1/3)\chi_{Pauli}$ Can measure χ_{Pauli} separately by NMR, Knight Shift Applies to "simple" metals: Na, Al, . . .
- 4. Localized systems, e.g. atoms
 - Consider effect of constant H field on electrons orbits around a center (the nucleus) $\nabla \times \mathbf{A} = \mathbf{H}$; can choose $A = -\frac{1}{2}\mathbf{r} \times \mathbf{H}$ Leads to diamagnetic + paramagnetic terms (see text)
 - Diamagnetism in closed shell system Larmor (Langevin) Diamagnetism $\chi = -(N/V)Z(e^2/6mc^2) < r^2 >$
 - Paramagnetic moments in open-shell systems Curie Law: Statistical Mechanics of independent moments in a magnetic field $\chi \approx (1/3)(N/V)(\mu_B^2/k_BT)p^2$, where $p \approx gJ(J+1)$ Brillouin function - see text
- 5. Origin of Paramagnetic moments: Electron-electron interactions beyond independent electron approximation

Hund's rules in atoms - due to electron-electron interactions Three rules: 1. Max S; 2. Max L consistent with 1; 3. J = L-S or L+S

- 6. Beyond independent electron models in solids: Atomic-like "local moments" in solids Rare Earth atoms - localized 4f states Transition metals - 3d states often act localized Next lectures: Kondo Effect in metals Interactions between different atoms leads to magnetic order
- 7. Extra include in there is time Van Vleck paramagnetism The nuclear spin adiabatic refrigerator