

Hyperfine structure (+ general "nuclear effects")

so far, we've examined:

- 1) gross structure - { electrostatic interactions
 Hydrogen-like \rightarrow just e^- charge + charge of nucleus
 more generally $\rightarrow e^-e^-$ interactions also plays a (shielding) big role

- 2) fine structure \rightarrow
 - relativistic corrections
 - spin-orbit interactions $\vec{s} \cdot \vec{L}$
 relates to orbital ang. mom.
 spin of electron
 $|\vec{\mu}| \sim \mu_B = \frac{e\hbar}{2me}$
 for single unpaired electron $H_{LS} = -\vec{\mu} \cdot \vec{B}$

recall $\vec{j} = \vec{L} + \vec{s}$
 for $s = \frac{1}{2}$, have $f = l = \frac{1}{2}$ and $j = l + \frac{1}{2}$
 if $l \neq 0$

- 3) hyperfine - shifts + energy splitting
 due to nuclear effects

(smaller than
 fine structure)

nucleus is not a point-like charge,

but has structure and spin

We'll treat energy corrections due to finite size, variations in the mass of nucleus shortly

shortly

② the hyperfine structure, in terms of

hyperfine energy levels, comes about due to the fact that the nucleus can have a spin angular momentum, and thus a magnetic moment.

We're going to try to give a rough sketch for how the nuclear spin gives rise to hyperfine structure. For a rigorous treatment based on a relativistic, quantum treatment, see lecture notes of N. Johnson

[https://www3.nd.edu/~johnson/
Publications/book.pdf](https://www3.nd.edu/~johnson/Publications/book.pdf)

Ch. #5

Let's start w/ the semi-classical picture as presented in Foot, which works well for hydrogen and will set the stage for a more general result.

the nucleus has a magnetic moment that can be related to the nuclear spin I as

$$\vec{\mu}_I = g_I \mu_N \vec{I}$$

Different conventions

Note: you'll see several different formulations, depending on the source. Some have positive g_I , some have negative.

Some absorb μ_N/μ_B factor into g_I , and express as

$$\vec{\mu}_I = g_I \mu_B \vec{I}$$

$$\text{where } \mu_N = \mu_B \frac{m_e}{m_p}, \text{ i.e.}$$

the nuclear magneton is lower than the Bohr magneton by the electron-to-proton mass ratio.

$$I, (m_I = -I, -I+1, \dots, I)$$

③ Given this nuclear magnetic moment $\vec{\mu}_I$, there is an energy correction ~~to~~ in the presence of a magnetic field (from general contribution) \vec{B}

$$H_{\text{HFS}} = -\vec{\mu}_I \cdot \vec{B} = -g_I \mu_N \vec{I} \cdot \vec{B}$$

while $g_S \approx 2$ for the electron spin

g_I value will depend on nuclear structure

In the absence of an external field, it is the effective field due to the presence of electrons that shifts the energy. There are ~~three~~ basic contributions to this field \vec{B} :

a) "field" produced by fact that electron has spin, and thus a magnetic moment. This can also be thought of in terms of the dipole-dipole interaction between $\vec{\mu}_I$ and $\vec{\mu}_e$ (when they're spatially separated), i.e.

this can also be recast as the field due to $\vec{\mu}_e$,

$$\frac{\mu_0}{4\pi r^3} [\vec{\mu}_I \cdot \vec{\mu}_e - 3(\vec{\mu}_I \cdot \hat{r})(\vec{\mu}_e \cdot \hat{r})]$$

$$\vec{r} \text{ where } \vec{r} = \vec{r}_e - \vec{r}_n \approx \vec{r}_e$$

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$$\begin{aligned} \vec{B}_{e,S} &= \frac{\mu_0}{4\pi} \left[\vec{\mu}_e - \frac{3(\vec{\mu}_e \cdot \hat{r})}{r^3} \hat{r} \right] \\ &= \frac{2\mu_0 \mu_B}{4\pi r^3} \left[\vec{s} - 3(\vec{s} \cdot \hat{r}) \hat{r} \right] \end{aligned}$$

Note: This is just the classical expression for point-like dipole, and we'll use an alternative approach for s-orbitals that overlap with the nucleus

b) the effective field that the nuclear spin feels due to the orbit of the electron. For states w/ $l \neq 0$, the "current" of the electron orbiting the nucleus leads to an effective field

$$\text{contribution} = -2 \frac{\mu_0 \mu_B}{4\pi r^3} \vec{l}$$

(c)

For the s -states, the contribution due to "dipole-dipole" averages to zero [anisotropic], but a Fermi contact interaction remains, which can be described in terms of the effective field $\vec{B}_{e,\text{Fermi}} = -\frac{2}{3} \mu_0 g_S \mu_B \vec{S} 8^3(r) \approx -\frac{2}{3} \mu_0 g_S \mu_B |\psi(r)|^2 \vec{S}$

④ So, summing up,

$$H_{HFS} = -\vec{\mu}_I \cdot \vec{B} \quad \text{and}$$

$$\vec{B}_e = -\frac{2\mu_0 M_B}{4\pi} \left[\frac{\vec{L}}{r^3} - \frac{(\vec{S} - (3\vec{S} \cdot \hat{r})\hat{r})}{r^3} + \frac{4\pi}{3} g_s |\psi(0)|^2 \vec{S} \right]$$

orbit term
dipole-dipole
Fermi contact

For an s-orbital, the first two contributions are both zero,
and we have for the $l=0$ state

$$H_{HFS} = \frac{2\mu_0 M_B g_I \mu_N}{4\pi 3} g_s |\psi(0)|^2 \vec{I} \cdot \vec{S}$$

for hydrogenic atoms $|\psi_{n0}(0)|^2 = \frac{1}{\pi} \left(\frac{Z}{na_0}\right)^3$
w/ s-orbital

$$\Rightarrow H_{HFS} = \frac{2}{3\pi} \mu_0 g_I g_s \mu_B \mu_N \frac{Z^3}{n^3 a_0^3} \vec{I} \cdot \vec{S}$$

for this $l=0$ s-state, $\vec{J} = \vec{S}$ and the electron spin and nuclear spin add together as $\vec{F} = \vec{I} + \vec{J}$ w/ F-values of I-J to I+J

~~so~~ $\vec{F} = \vec{I} + \vec{S}$

recall our treatment of $\vec{L} \cdot \vec{S}$ w/ $\vec{J} = \vec{L} + \vec{S}$

and $J-I$ to $J+I$ for $J>I$

$$\Rightarrow \vec{F}^2 = \vec{I}^2 + \vec{S}^2 + 2\vec{I} \cdot \vec{S}$$

note: mf values from $-F, -F+1, \dots, F-1, F$
expectation values

$$\therefore \vec{I} \cdot \vec{S} = \frac{1}{2} [\vec{F}^2 - \vec{I}^2 - \vec{S}^2] \rightarrow [F(F+1) - I(I+1) - S(S+1)]$$

ANS

⑤

this gives us an energy correction to hydrogenic s-state levels of

$$\Delta E = \frac{g_I}{3\pi} \mu_0 g_I g_S \mu_B \mu_N \frac{Z^3}{n^3 a_0^3} \left[\frac{F(F+1) - I(I+1) - S(S+1)}{2} \right]$$

we expect that this should work fairly well for hydrogen & hydrogenic atoms, so let's check.

$$Z=1$$

for the hydrogen nucleus (i.e. a proton), $g_I = 5.5883$
and $I = \frac{1}{2}$

for the electron we have $S = \frac{1}{2}$ and $g_S \approx 2$ (we'll look at $n=1$)

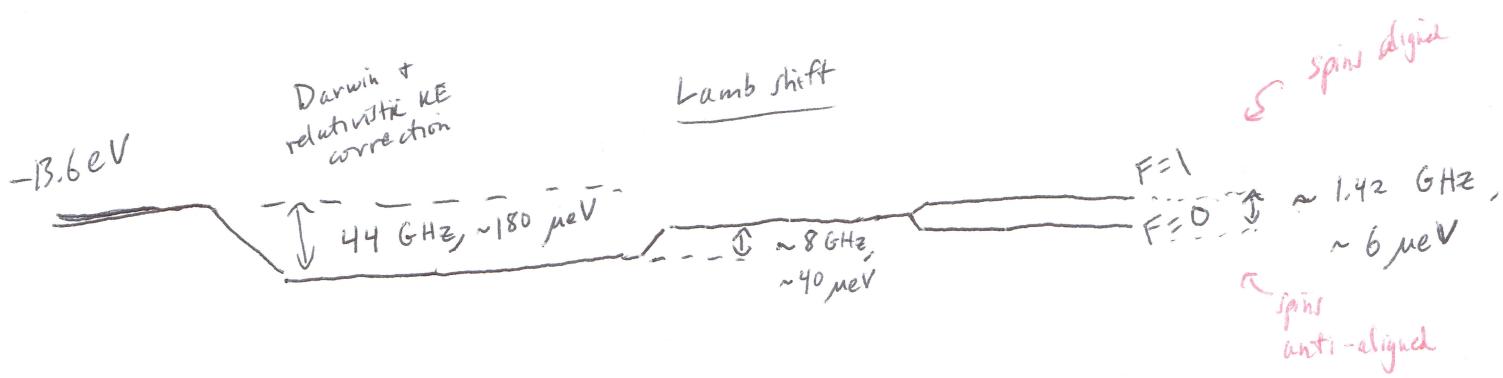
The allowed F-values are $F = 0, 1$.

$$\Delta E_{F=0} = -\frac{3}{2} \left[\frac{2}{3\pi} (5.5883) \mu_0 \mu_B \mu_N \frac{1}{a_0^3} \right] \quad \left| \begin{array}{l} \Delta E_{F=0} \\ \hline \Delta E_{F=1} \end{array} \right| = 3$$

$$\Delta E_{F=1} = +\frac{1}{2} \left[\frac{2}{3\pi} \quad " \quad " \quad \right]$$

$$\Delta E_{F=1} - \Delta E_{F=0} = \frac{4}{3\pi} g_I \mu_0 \mu_B \mu_N \frac{1}{a_0^3} \approx h \times 1.42 \text{ GHz}$$

this corresponds to $\lambda = 21 \text{ cm}$, used extensively in radio astronomy



⑥ This result fails for the more general case (multi-electron atom, $l \neq 0$)
but a result having similar form follows from a proper treatment using relativistic quantum mechanics.

$$H_{\text{HFS}} = H_{\text{dip}} + H_{\text{quad}}$$

\uparrow \uparrow
magnetic electric
dipole moment quadrupole
of nucleus moment of
 nucleus

(relevant for $I > \frac{1}{2}$
and $l \neq 0$, interaction of quadrupole
moment w/ $\vec{\nabla}E$)

$$H_{\text{dip}} = A_{\text{HFS}} \vec{I} \cdot \vec{J}$$

\uparrow
same form as
before

roughly
we calculated A for hydrogen-1, He atoms, but in general these values come from experiment.

$$H_{\text{quad}} = B_{\text{HFS}} \frac{3(\vec{I} \cdot \vec{J})^2 + \frac{3}{2}(\vec{I} \cdot \vec{J}) - I(I+1)J(J+1)}{2IJ(2I-1)(2J-1)}$$

{ only for
 $I = \frac{1}{2}, l \neq 0$ }

F is a good quantum number here, as is its z-projection, m_F .
 $(\vec{F} = \vec{I} + \vec{J})$ \hookrightarrow will be a good Q in B-fields

These $|I, J, F, m_F\rangle$ states ($|F, m_F\rangle$ for short) can be expressed in terms of the quantum #s of the nuclear spin & electron spin as.

$$|F, m_F\rangle = \sum_{m_I, m_J}$$

satisfying
 $m_F = m_I + m_J$

$\boxed{\langle F, m_F | I, m_I, J, m_J \rangle} |I, m_I, J, m_J \rangle$
 ~~$\langle I, m_I, J, m_J | F, m_F \rangle$~~ ~~$|I, m_I, J, m_J \rangle$~~

\hookrightarrow Clebsch-Gordan coefficient

$$\langle I, m_I, J, m_J | F, m_F \rangle = \sqrt{2F+1} (-1)^{I-J+m_F} \begin{pmatrix} I & J & F \\ m_I & m_J & -m_F \end{pmatrix}$$

Wigner 3j

⑦ The first-order energy shift is given by

$$\Delta E_{HFS} = \frac{1}{2} A_{HFS} K + B_{HFS} \frac{\frac{3}{2} K(K+1) - 2I(I+1)J(J+1)}{2I(2I-1)2J(2J-1)}$$

$$\text{where } K = F(F+1) - I(I+1) - J(J+1) = 2\langle \vec{I} \cdot \vec{J} \rangle$$

see slides for some experimental values.

for some details on measurement
of A_{HFS} & B_{HFS} constants,
see RMP, 49 31 (1997)
by Arimondo & Inguscio

Some notes : $A_{HFS} > 0$ is referred to as "regular"

whereas

$A_{HFS} < 0$, as in ^{40}K , is called "inverted"

For the alkalis, $S = \frac{1}{2}$, the ground state ($\ell=0$) Zeeman levels, i.e. the states $|F, m_F\rangle$ are constructed by superposition of only 2 states in the $|I, m_I, J, m_J\rangle$ basis.

For $\ell=0$, we have $J=5$ such that we use $|I, m_I, S, m_S\rangle$, and call these states $|m_S, m_I\rangle$ for short.

recall that $m_F = m_S + m_I$, and $m_S = \pm \frac{1}{2}$, such that the allowed m_I values are $m_I = m_F - m_S = m_F \mp \frac{1}{2}$ for $m_S = \pm \frac{1}{2}$

In general, $|F, m_F\rangle = \underbrace{\alpha |m_S = \frac{1}{2}, m_I = m_F - \frac{1}{2}\rangle + \beta |m_S = -\frac{1}{2}, m_I = m_F + \frac{1}{2}\rangle}_{\alpha, \beta \text{ are Clebsch-Gordan coefficients.}}$ This form will help us understand how we have states w/ vanishing magnetic moment ($m_F=0$) at low fields.

(8)

Other "nuclear" effects

We also get corrections to the energy levels due to details of the nucleus' mass, variations in its spin, and the non-zero size of the nucleus (not a point-like charge).

- ① Size effects → Volume shift - in HW1, you calculated the energy corrections that one would expect due to the ~~nucleus~~ deviation of the nuclear charge distribution from that of a point charge, which led to a potential that was not proportional to $\frac{1}{r}$ for small r values
- This effect is sometimes considered a hyperfine effect (comes from nuclear property) and sometimes absorbed as part of other effects (as in the discussion of proton size puzzle w/ Lamb shift measurement)

Simple toy model
(~ same as HW#1)
for hydrogen-like atom

Consider a sphere w/ uniform charge density, having total charge $Q_N = Ze$ and radius $R \approx A^{1/3} r_0$

where $A \approx 2Z$ is the atomic mass number (# of nucleons)
and $r_0 = 1.2 \text{ fm}$ (radius close to proton radius $r_p \approx 0.9$)

The potential energy experienced by the electron

$$V(r) = \begin{cases} \frac{Ze^2}{4\pi\epsilon_0 R} \frac{1}{2} \left(\frac{r^2}{R^2} - 3 \right) & \text{for } r \leq R \\ \frac{Ze^2}{4\pi\epsilon_0 r} & \text{for } r > R \end{cases}$$

④

this volume effect leads to an energy correction

$$\Delta E = \langle H' \rangle \quad \text{where} \quad H' = V(r) - V_{\text{point charge}}$$

$$= \begin{cases} \frac{Ze^2}{4\pi\epsilon_0 R} \frac{1}{2} \left(\frac{r^2}{R^2} + \frac{2R}{r} - 3 \right) & \text{for } r \leq R \\ 0 & \text{for } r > R \end{cases}$$

$$\Delta E = \langle H' \rangle = \langle \psi_{\text{atom}} | H' | \psi_{\text{atom}} \rangle$$

because R is so small, this effect
is only important (non-zero) for
 $\ell = 0$ states

$$\Delta E = \frac{Ze^2}{4\pi\epsilon_0 2R} \int_0^R |R_{\text{atom}}(r)|^2 \left(\frac{r^2}{R^2} + \frac{2R}{r} - 3 \right) r^2 dr$$

$$\Delta E = \frac{Ze^2}{4\pi\epsilon_0 2R} |R_{n0}(0)|^2 \underbrace{\int_0^R \left(\frac{r^2}{R^2} + \frac{2R}{r} - 3 \right) r^2 dr}_{\frac{R^3}{5}}$$

$$= \frac{Ze^2}{10\epsilon_0} R^2 |R_{n0}(0)|^2 \rightarrow \frac{4Z^3}{h^3 a_0^3}$$

$$= \frac{Z^4}{a_0^3 h^3} \frac{e^2 R^2}{10\epsilon_0} \Rightarrow \text{Foot Eq (6.26)}$$

↳ result applied to alkalis, etc.
based on central field approximation

$$\cancel{\frac{Z^4}{a_0^3 h^3}} \cancel{\frac{e^2 R^2}{10\epsilon_0}}$$

(1D)

② mass effect → we'll only discuss the reduced mass effect, ~~due to~~

which requires a correction due to our assuming that $\frac{M_{\text{nuclear}}}{m_e} = \infty$

$$E_n = -\frac{e^2 \cancel{\mu}}{4\pi \epsilon_0 a_0} \frac{Z^2}{2n^2} = \frac{-e^2}{4\pi \epsilon_0 a_0} \frac{Z^2}{2n^2} \frac{M_N}{m_e + M_N} = \frac{E_n^\infty}{\frac{m_e}{M_N} + 1} = E_n^\infty \left(1 + \frac{m_e}{M_N}\right)^{-1}$$

$$\approx E_n^\infty \left(1 - \frac{m_e}{M_N}\right)$$

so, the correction for a finite nuclear mass

$$\text{ii } \Delta E_n = -E_n^\infty \frac{m_e}{M_N}$$

this correction always $\propto -E_n^\infty$, such that it shifts energies up, and correction term gets smaller as M_N increases.